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- (71) Applicant: THE PENN STATE RESEARCH FOUNDATION [US/US]; 304 Old Main, University Park, PA 16802 (US).
- (72) Inventors: ZHANG, Xumu; 276 Camelot Lane, State College, PA 16803 (US). XIAO, Dengming; 2201 14th Avenue, Apt. #4-204, Longmont, CO 80501 (US).
- (74) Agent: MONAHAN, Thomas, J.; Intellectual Property Office, The Pennsylvania State University, 113 Technology Center, University Park, PA 16802-7000 (US).
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(54) Title: CHIRAL LIGANDS, TRANSITION-METAL COMPLEXES THEREOF AND USES THEREOF IN ASYMMETRIC REACTIONS

(57) Abstract: Chiral ligands and transition metal complexes based on such chiral ligands useful in asymmetric catalysis are disclosed. The chiral ligands include phospholanes, P, N ligands, N, N ligands, biphenols, and chelating phosphines. The ferrocene-based iridium (*R,R*)-f-binaphane complex reduces imines to the corresponding amines with 95-99.6 % enantioselectivity and reduces β -substituted- α -arylenamides with 95 % enantioselectivity. The transition metal complexes of the chiral ligands are useful in asymmetric reactions such as asymmetric hydrogenation of imines, asymmetric hydride transfer reactions, hydrosilylation, hydroboration, hydrovinylation, hydroformylation, allylic alkylation, cyclopropanation, Diels-Alder reaction, Heck reaction, isomerization, Aldol reaction, Michael addition and epoxidation reactions.

CHIRAL LIGANDS, TRANSITION-METAL COMPLEXES THEREOF
AND USES THEREOF IN ASYMMETRIC REACTIONS

5

BACKGROUND OF THE INVENTION

1. FIELD OF THE INVENTION

10 The present invention relates to chiral ligands and
transition metal complexes thereof that are useful in
asymmetric reactions. More particularly, the present
invention relates to chiral phospholanes, P,N ligands,
N,N ligands, biphenols, and chelating phosphines and
15 transition metal complexes thereof that are useful in
asymmetric catalysis.

2. DESCRIPTION OF THE PRIOR ART

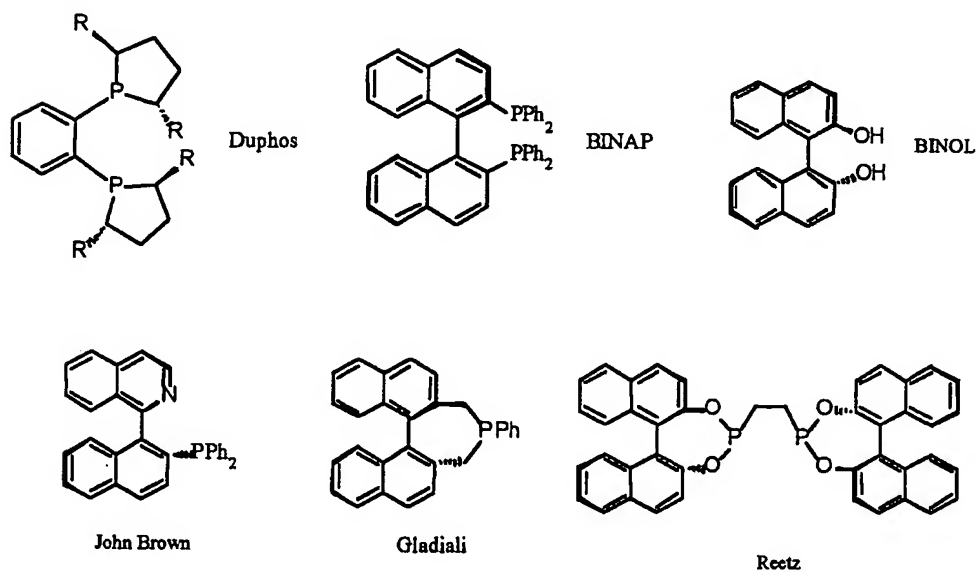
20 Discovery of new chiral ligands has been an
essential element in the development of highly
enantioselective transition metal-catalyzed reactions.
New structural motifs play an important role in
dictating enantioselectivities and reactivities of a
25 reaction.

With the growing demand of enantiomerically pure
compounds in pharmaceutical and agrochemical industry,
asymmetric catalysis has become increasingly more
30 important because of its high efficiency.

- For example, biaryl atropisomeric ligands have been explored as effective ligand scaffolds for many asymmetric transformations. One of the most frequently used chiral chelating phosphines is BINAP (Noyori, R.;
- 5 Takaya, H. *Acc. Chem. Res.* **1990**, *23*, 345, Ohkuma, T.; Koizumi, M.; Doucet, H.; Pham, T.; Kozawa, M.; Murata, K.; Katayama, E.; Yokozawa, T.; Ikariya, T.; Noyori, R. *J. Am. Chem. Soc.* **1998**, *120*, 13529.).
- 10 Another family of excellent chiral phosphines is so called DuPhos (Burk, US patent 5,329,015, US Patent 5,202,493, US Patent 5,329,015, Burk, M, J. *J. Am. Chem. Soc.* **1991**, *113*, 8518, Burk, M. J.; Feaster, J. E.; Nugent, W. A.; Harlow, R. L. *J. Am. Chem. Soc.* **1993**,
- 15 *115*, 10125. Burk, M. J.; Wang, Y. M.; Lee, J. R. *J. Am. Chem. Soc.* **1996**, *118*, 5142), which has a rigid 1, 2-bis(phosphino)benzene backbone and electron-donating phospholane groups.
- 20 Gladiali et al. (Gladiali, S.; Dore, A.; Fabbri, D.; Lucchi, O. D.; Manassero, M. *Tetrahedron Asymmetry*, **1994**, 511.) made monodentate chiral phospholanes bearing the 1, 1'-binaphthyl framework. However the method for their synthesis is not feasible to make the
- 25 corresponding chelating chiral phospholanes. Stelzer et al. (Bitterer, F.; Herd, O.; Kuhnelt, M.; Stelzer, O.; Weferling, N.; Sheldrick, W. S.; Hahu, J.; Nagel, S.; Rosch, N. *Inorg. Chem.* **1998**, *37*, 6408) only made racemic chelating phospholanes.

Reetz et al. prepared chelating chiral phosphinites using readily accessible binaphthanols as starting materials and demonstrated that they are excellent ligands for Rh-catalyzed asymmetric hydrogenation of dehydroaminoacids (Reetz, M. T.; Gosberg, A.; Goddard, R.; Kyung, S. *J. Chem. Soc., Chem. Commun.* **1998**, 2077). John Brown made a chiral phosphine and pyridine ligand with a biaryl chirality. Several related chiral ligands are shown in the Figure below.

10

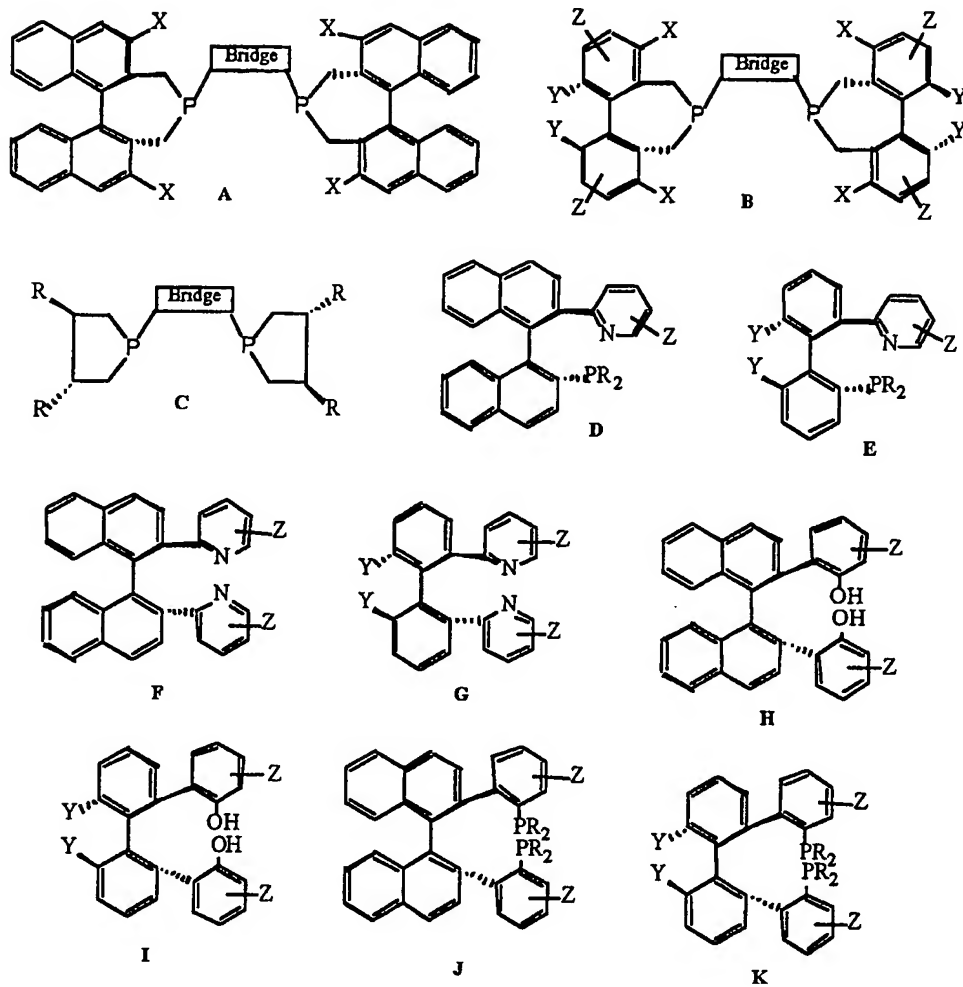


While these ligands have been useful in a number of asymmetric reactions, there are still many more asymmetric transformations that can benefit from the discovery of new chiral ligands.

SUMMARY OF INVENTION

The present invention includes a ligand selected from the group consisting of compounds represented by A through K:

5



W

herein the bridge group is selected from the group

consisting of: $(CH_2)_n$ wherein n is an integer ranging from 1 to 8, $(CH_2)_nW(CH_2)_m$ wherein n and m are independently an integer ranging from 1 to 8 and W , wherein W is a divalent group selected from the group consisting of: 1,2-divalent phenyl, 2,2'-divalent 1,1'-biphenyl, 2,2'-divalent-1,1'-binaphthyl, ferrocene, and a substituted derivative thereof; wherein each substituent in said substituted derivative is selected from the group consisting of: aryl, alkyl having 1-8 carbon atoms, F, Cl, Br, I, COOR, SO₃R, PR₃R₂, OR, SR, PR₂, AsR₂, SbR₂, aryloxy, nitro, NR₂, vinyl, substituted vinyl and a combination thereof, wherein each R is independently selected from the group consisting of: hydrogen, alkyl, aryl, alkaryl and aralkyl; wherein each X is independently selected from the group consisting of: hydrogen, halide, alkyl, aryl, alkoxy, silane, carboxylate and amide; each Y is independently selected from the group consisting of: hydrogen, alkyl, aryl, alkoxy, carboxylate and amide; and each Z is independently selected from the group consisting of: hydrogen, alkyl, aryl, alkoxy, amide, carboxylate, and a heterocyclic compound.

The present invention further includes a catalyst prepared by a process comprising contacting a transition metal salt, or a complex thereof, and a ligand selected from the group consisting of compounds represented by A through K as described above.

The present invention still further includes a process for preparation of an asymmetric compound using

a catalyst according to the present invention. The process comprises contacting a substrate capable of forming an asymmetric product by an asymmetric reaction and a catalyst prepared by a process comprising
5 contacting a transition metal salt, or a complex thereof, and a ligand selected from compounds represented by A through K as described above.

The ferrocene-based irridium (*R,R*)-f-binaphane
10 complex reduces imines to the corresponding amines with 95-99.6% enantioselectivity and reduces β -substituted- α -arylenamides with 95% enantioselectivity.

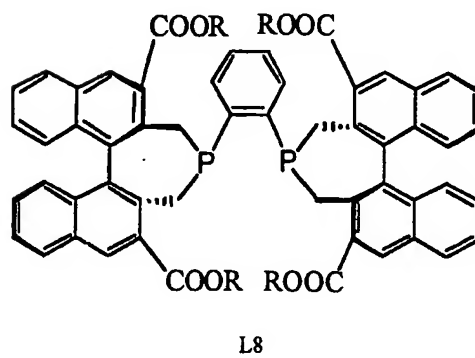
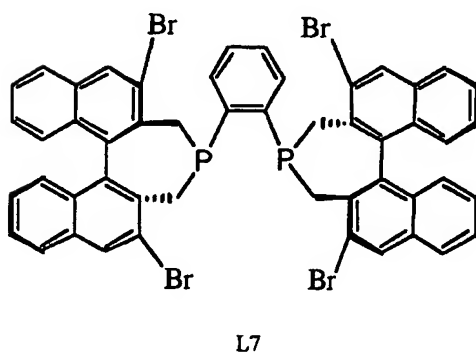
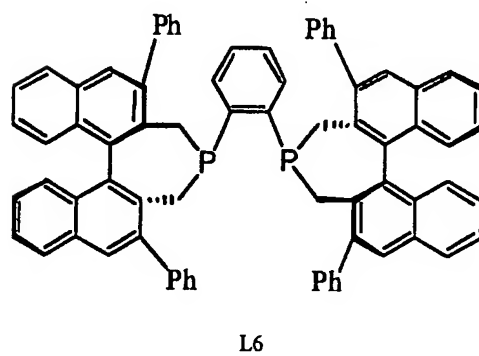
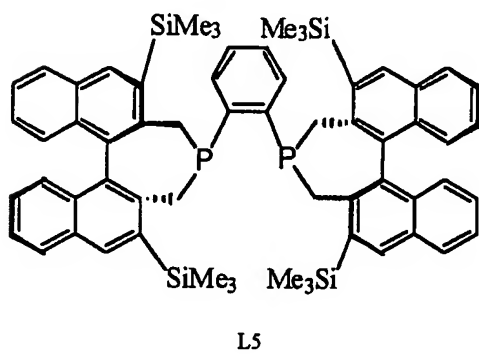
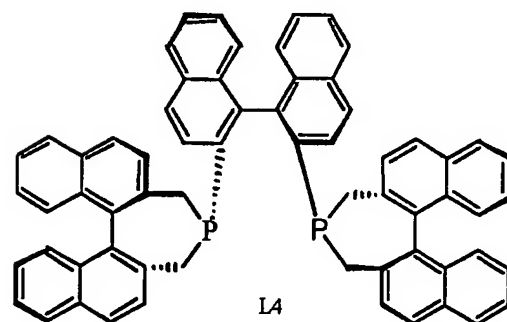
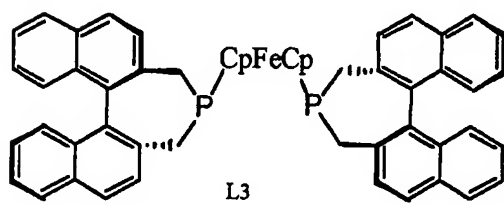
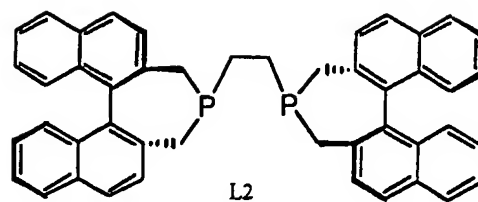
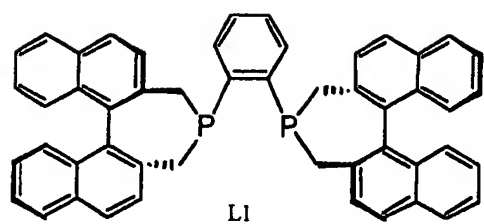
DETAILED DESCRIPTION OF INVENTION

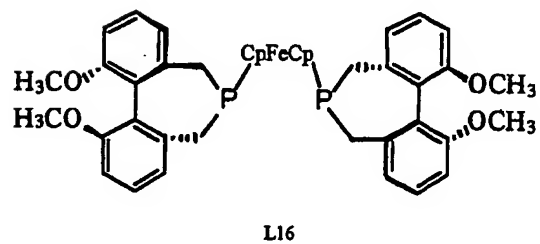
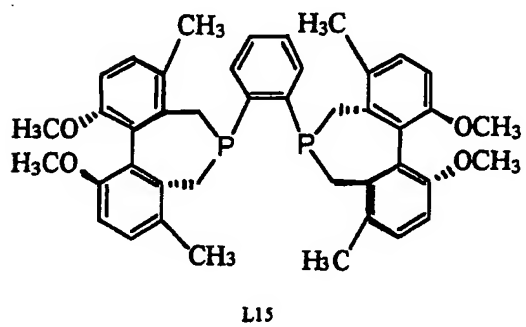
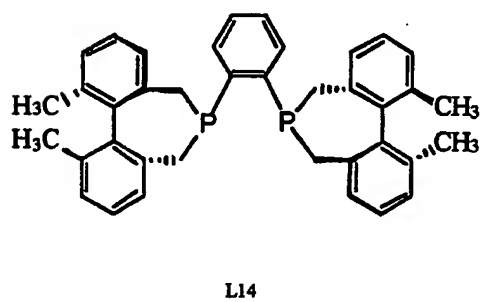
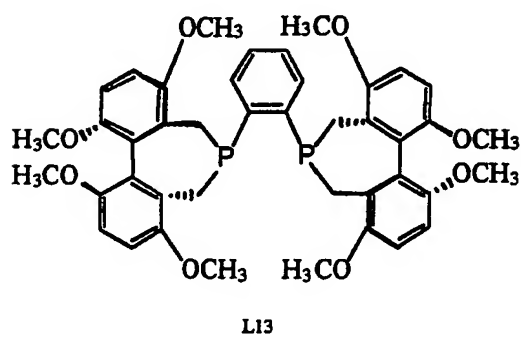
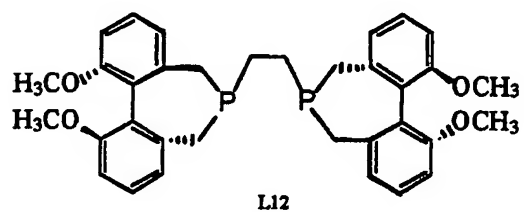
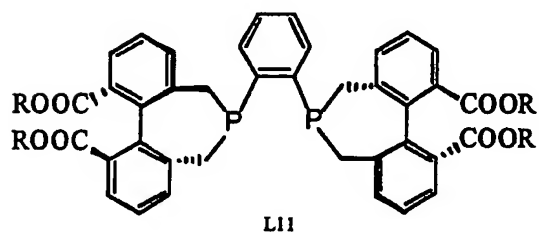
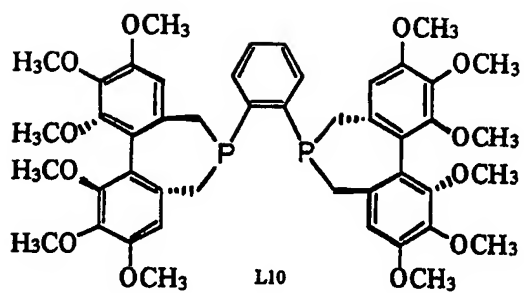
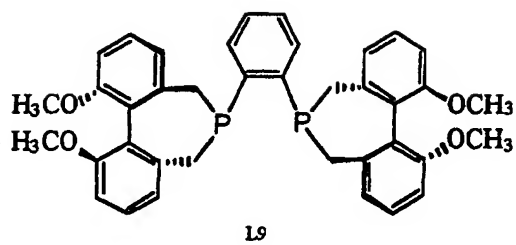
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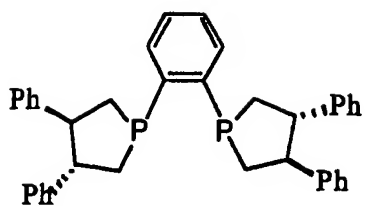
The present invention includes new phospholane ligands with mixed biaryl chirality. The P,N ligands, N,N ligands, biphenols and chelating phosphines are also derivatives of biaryl atropisomers. Also included are
20 chiral five-membered ring phospholanes with stereogenic centers in 3,4 positions, phospholanes with a chiral biaryl atropisomer as the backbone and atropisomers of P,N ligands, N,N ligands, biphenols and chelating bisphophines. These chiral ligands can be used to
25 facilitate a variety of metal-catalyzed asymmetric transformations. The bridge group can be $(CH_2)_n$ wherein n is an integer ranging from 1 to 8, $(CH_2)_nW(CH_2)_m$ wherein n and m are independently an integer ranging from 1 to 8 and W , wherein W is a divalent group
30 selected from the group consisting of: 1,2-divalent phenyl, 2,2'-divalent 1,1'-biphenyl, 2,2'-divalent-1,1'-

binaphthyl, ferrocene, and a substituted derivative thereof. Each substituent in the substituted derivative can be aryl, alkyl having 1-8 carbon atoms, F, Cl, Br, I, COOR, SO₃R, PR₃R₂, OR, SR, PR₂, AsR₂, SbR₂, aryloxy, 5 nitro, NR₂, vinyl, substituted vinyl and a combination thereof and each R can independently be hydrogen, alkyl, aryl, alkaryl and aralkyl. Each X can independently be hydrogen, halide, alkyl, aryl, alkoxy, silane, carboxylate and amide, each Y can independently be 10 hydrogen, alkyl, aryl, alkoxy, carboxylate and amide and each Z can independently be hydrogen, alkyl, aryl, alkoxy, amide, carboxylate, and a heterocyclic compound (i.e., a nitrogen, sulfur or oxygen heterocycle).

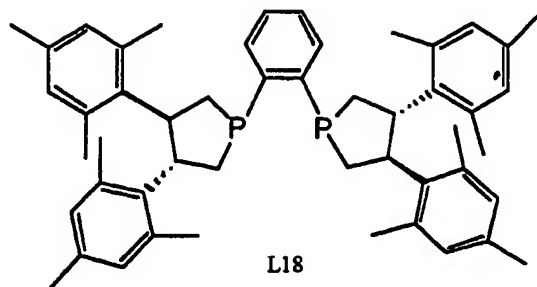
15 For each class of A to K ligands, the corresponding enantiomer, as well as enantiomeric mixtures, are also contemplated. A and B ligands are chelating chiral phospholanes with biaryl chirality in their backbone. C 20 ligands have five-membered ring phospholanes with stereogenic centers in 3,4 positions. D and E are chiral P,N ligands with biaryl chirality. F and G are chiral N,N ligands with biaryl chirality. H and I are chiral biphenols with biaryl chirality. J and K are chiral phosphines with biaryl chirality. The preferred 25 ligands of the present invention are selected from ligands designated A through K, which include members represented by the formula L1 through L56 depicted below:



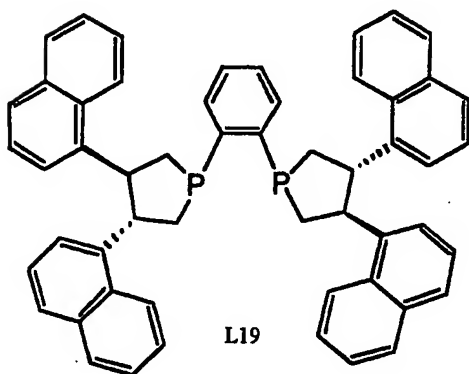




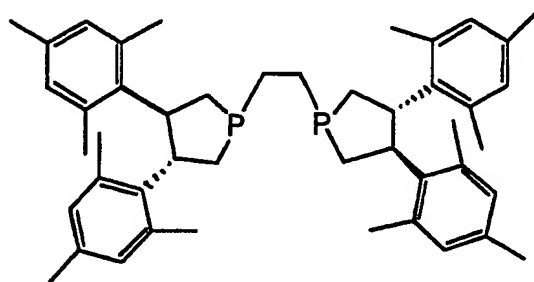
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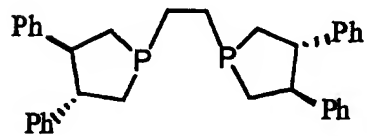
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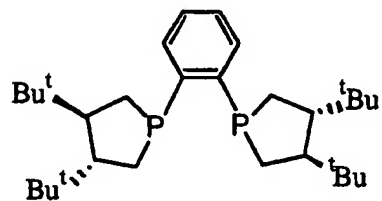
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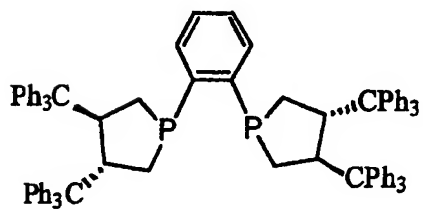
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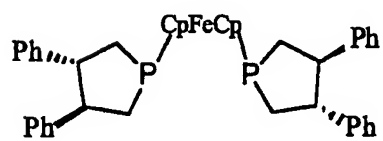
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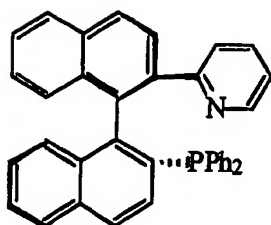


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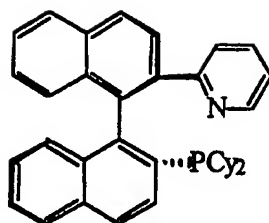


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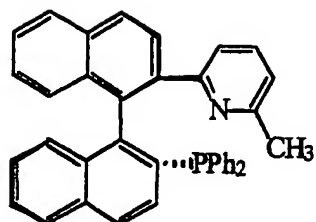
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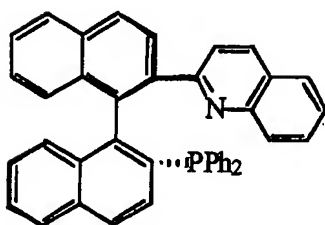
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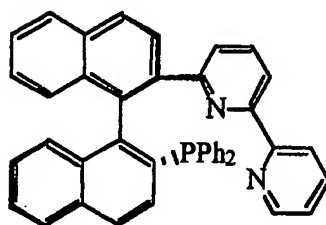
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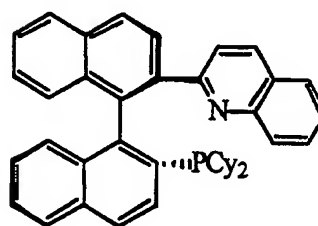
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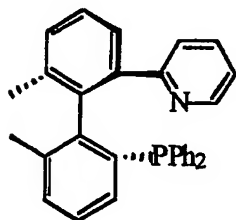
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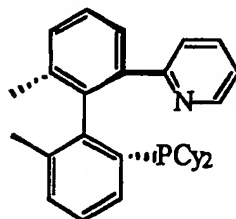
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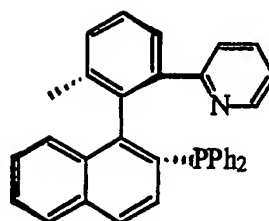
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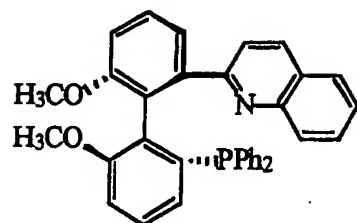
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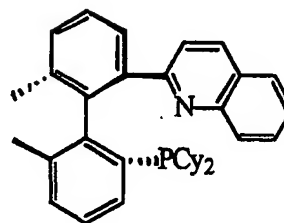
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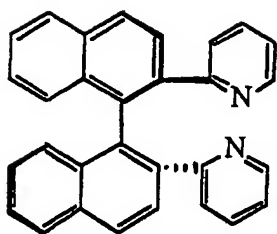
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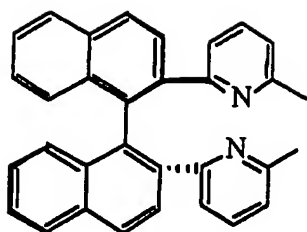
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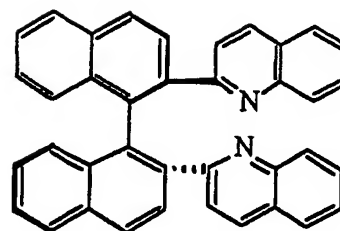
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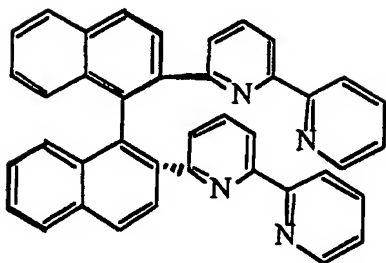
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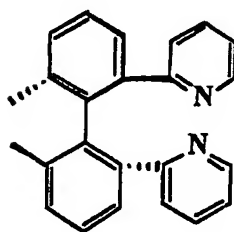
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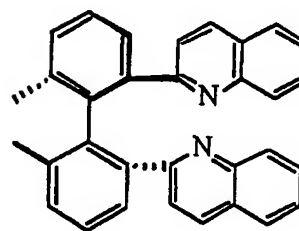
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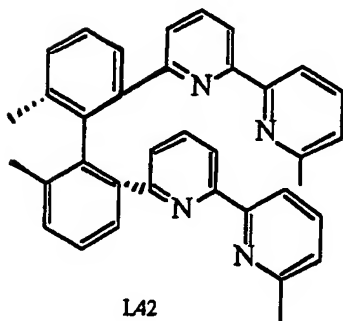
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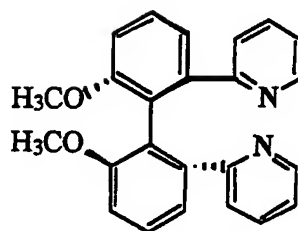
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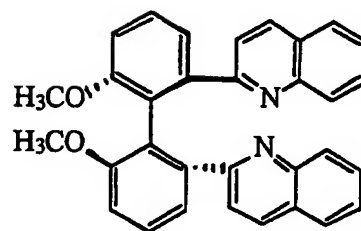
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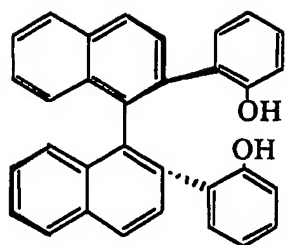
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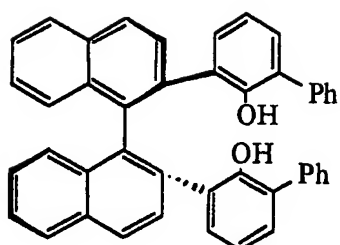
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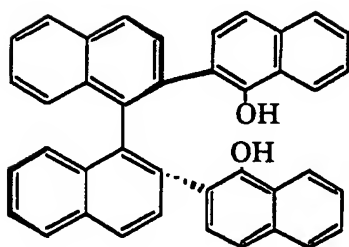
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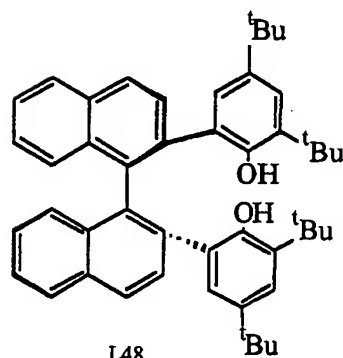
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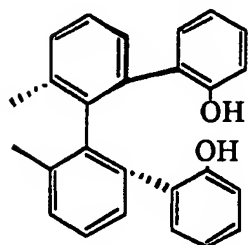
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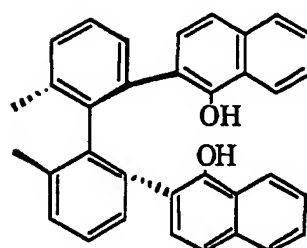
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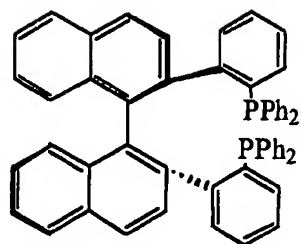
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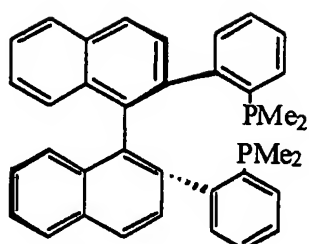
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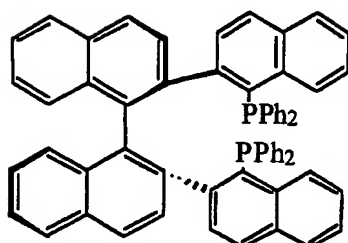
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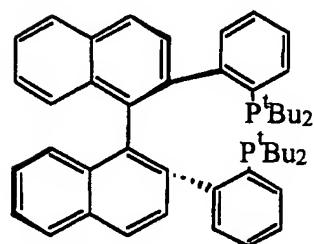
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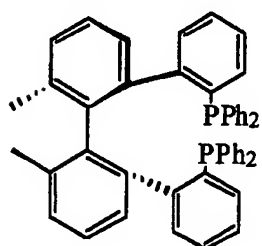
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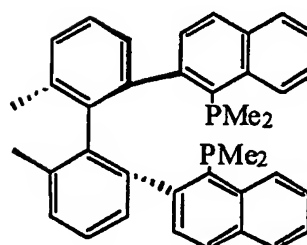
L53



L54



L55

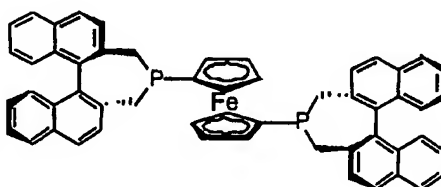


L56

L1 to L8 are examples of A ligands. L9 to L16 are examples of B ligands. L17 to L24 are examples of C ligands. L25 to L35 are examples of D and E ligands. L36 to L44 are examples of F and G ligands. L45 to L50
5 are examples of F and G ligands. L51 to L56 are examples of J and K ligands.

f-Binaphane ligand and transition metal complexes thereof are preferred, irridium complexes of f-binaphane
10 being the most preferred. The unsubstituted (R,R)-f-binaphane ligand is represented by the formula:

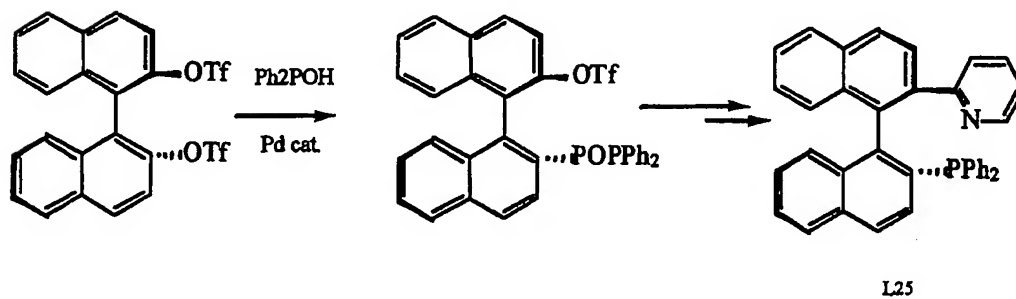
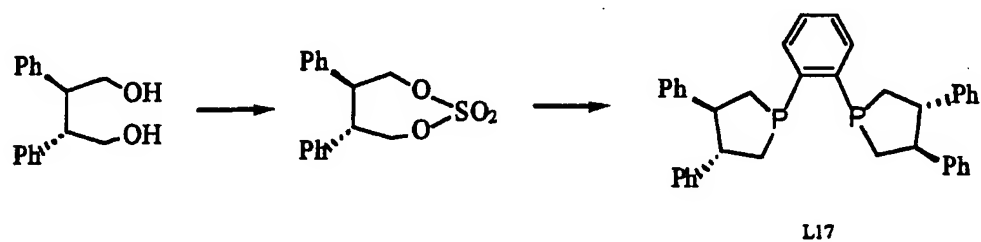
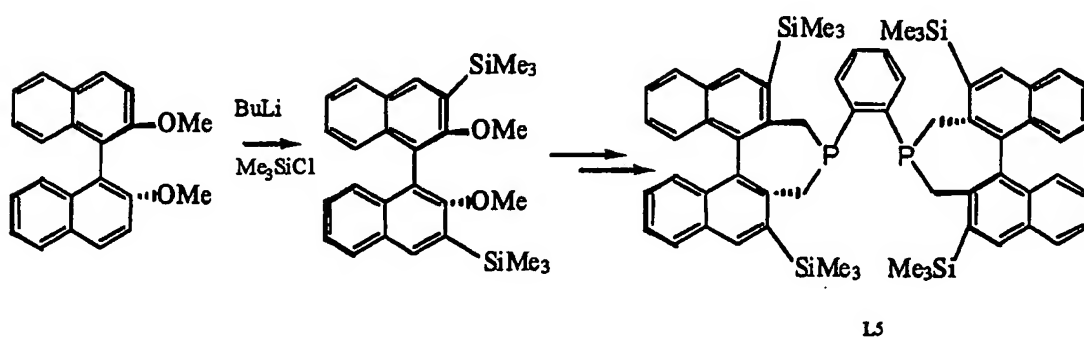
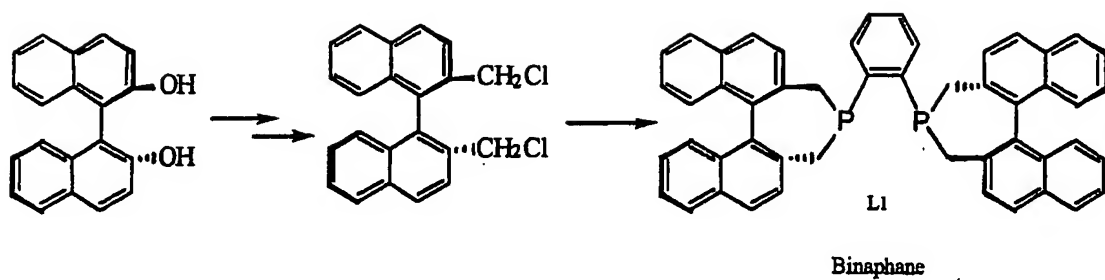
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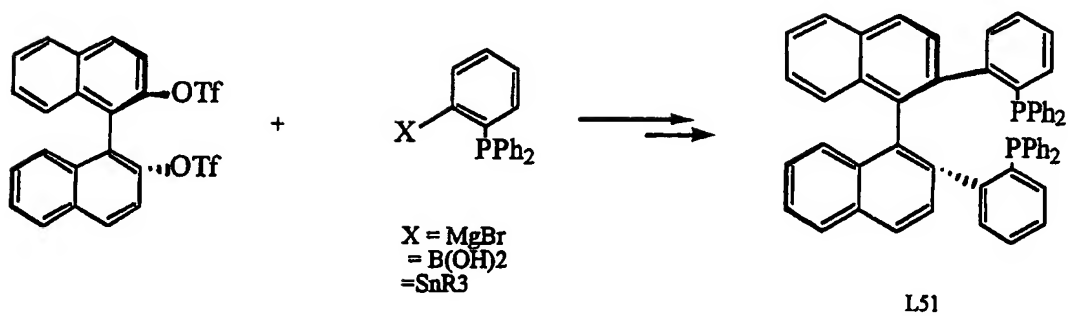
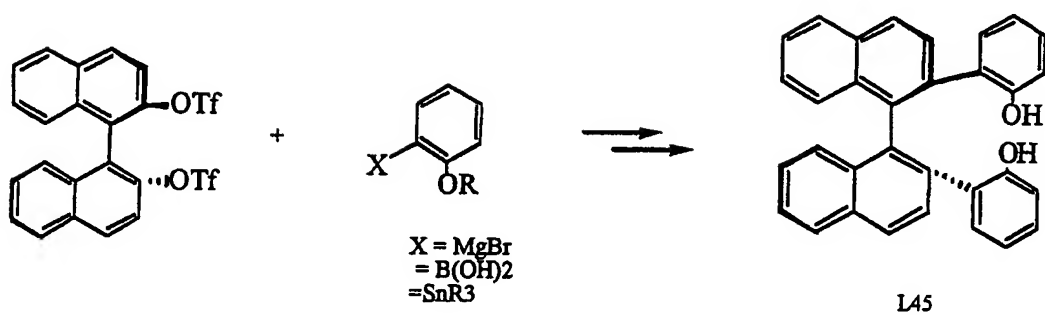
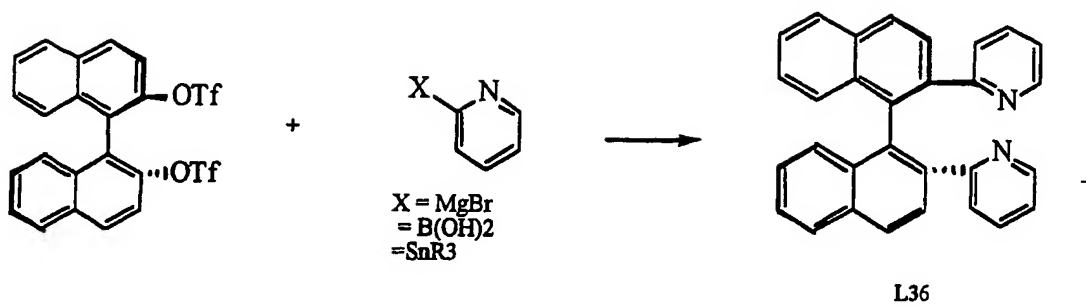


20

The highest enantioselectivity (>99%ee) has been achieved in the asymmetric hydrogenation of imines using
25 Ir-f-binaphane complex as the catalyst.

The preparation of L1, L5, L17, L25, L36, L45, L51 and (R,R)-binaphane are illustrated below. Other members of L1 to L56 ligands can be prepared by similar
30 procedures.

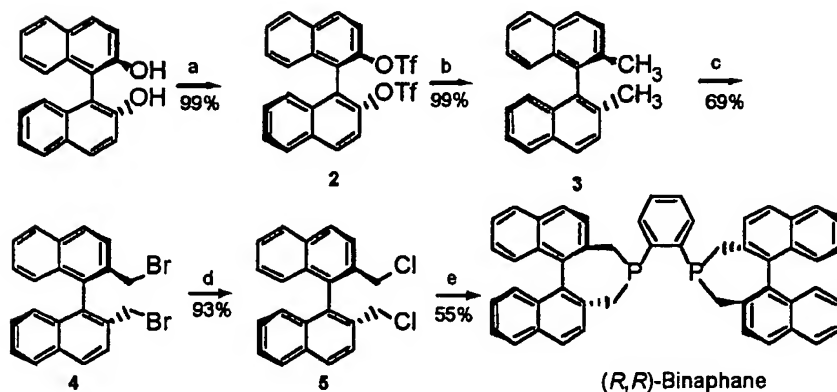




The ligand according to the present invention can be racemic, i.e., racemic mixture of enantiomers, or a non-racemic mixture of enantiomers. Preferably, the ligand according to the present invention is one of the enantiomers. When the ligand is a non-racemic mixture of enantiomers, preferably it has an optical purity of at least 85% ee, more preferably, it has an optical purity of at least 95% ee.

According to the above reaction scheme, (*R,R*)-1,2-bis((*R*)-4,5-dihydro-3H-dinaphtho[2,1-*c*:1',2'-*e*]phosphenino)benzene, abbreviated as (*R,R*)-binaphane, has been prepared in high yield and high optical purity. This chiral chelating phosphine has a rigid 1,2-bis(phosphino)benzene backbone and has both binaphthyl chirality and a phospholane functionality.

The preparation procedure is illustrated below:



^a $\text{TiF}_4 \cdot \text{OEt}_2$, Py, CH_2Cl_2 . ^b MeMgBr , $\text{NiCl}_2(\text{dppp})$, Et_2O . ^c NBS, Benzoyl peroxide, CCl_4 . ^d UCl_4 , DMF. ^e 1,2-Bisphosphinobenzene, NaH,

As illustrated schematically above, (R,R)-binaphane was prepared employing a practical synthesis route based on readily accessible starting materials.

- 5 Enantiomerically pure binaphthol can be easily obtained using a classic resolution procedure (Cai, D.; Hughes, D. L.; Verhoever, T. R.; Reider, P. J. *Tetrahedron Letter*, 1995, 7991). (R)-2,2'-bistriflate-1,1'-binaphthyl (2) was made from (R)-binaphthol by treating
10 with excess triflic anhydride and pyridine in CH₂Cl₂. Kumada coupling of bistriflate (2) with methyl magnesium bromide gave (R)-2,2'-dimethyl-1,1'-binaphthyl (3) in high yield. (R)-2,2'-Dibromomethyl-1,1'-binaphthyl (4) was prepared by bromination of 3 with NBS. A simple
15 anion exchange of (R)-2,2'-dibromomethyl-1,1'-binaphthyl (4) with LiCl afforded (R)-2,2'-dichloromethyl-1,1'-binaphthyl (5) in high yield. A key element of our synthesis of chelating phospholane such as binaphane is utilization of a less reactive (R)-2,2'-dichloromethyl-
20 1, 1'-binaphthyl (5) to avoid the intermolecular reaction with phosphine anions which existed when using a more reactive (R)-2,2'-dibromomethyl-1,1'-binaphthyl (4). Refluxing (R)-2,2'-dichloromethyl-1,1'-binaphthyl (5) with 1,2-bis(phosphino)benzene and NaH in THF
25 followed by recrystallization from ether gave (R,R)-binaphane in 55% yield. This efficient synthesis allows us to make binaphane in a large scale. Using this procedure, we have also made the corresponding monodentate chiral binaphthyl phospholane from
30 phenylphosphine in >90% yield.

In a similar connectivity as in binaphane, new chiral five-membered ring chelating phospholanes with stereogenic centers in 3,4 positions can be effective. An important transformation in making the binaphthyl phospholane ligand is Kumada coupling of ArOTf with RMgBr (from 2 to 3). Stille and Suzuki coupling may also work as well. Based on these coupling strategies, a series of new atropisomers of P,N ligands, N,N ligands, biphenols and chelating bisphosphines can be derived.

Chiral f-binaphane (or a substituted derivative thereof) was prepared according to a process comprising contacting chiral 1,1'-di(chloromethyl)binaphthyl (or a substituted derivative thereof) and 1,1'-bisphosphinoferrocene (or a substituted derivative thereof) in the presence of a base and a solvent to produce f-binaphane (or the substituted derivative thereof). Preferably, the base is NaH and the solvent is tetrahydrofuran (THF).

The present invention also includes a catalyst prepared by a process comprising contacting a transition metal salt, or a complex thereof, and a ligand selected from the group consisting of compounds represented by A through K.

As for the ligand, the catalyst according to the present invention can be racemic, such as, a racemic mixture of enantiomers, or it can be a non-racemic mixture of enantiomers. Preferably, the catalyst

according to the present invention is one of the enantiomers. When the ligand according to the present invention is a non-racemic mixture of enantiomers, preferably it has an optical purity of at least 85% ee, more preferably, it has an optical purity of at least 95% ee.

Suitable transition metals for the preparation of the catalyst include Pt, Pd, Rh, Ru, Ir, Cu, Ni, Mo, Ti, V, Re and Mn.

The catalyst can be prepared by contacting a transition metal salt or its complex and a ligand selected from A through K. The transition metal salt or complex can be PtCl_2 ; $\text{Pd}_2(\text{DBA})_3$; $\text{Pd}(\text{OAc})_2$; $\text{PdCl}_2(\text{RCN})_2$; $[\text{Pd}(\text{allyl})\text{Cl}]_2$; $[\text{Rh}(\text{COD})\text{Cl}]_2$; $[\text{Rh}(\text{COD})_2]\text{X}$; $\text{Rh}(\text{acac})(\text{CO})_2$; $\text{Rh}(\text{ethylene})_2(\text{acac})$; $\text{Rh}(\text{CO})_2\text{Cl}_2$; $\text{Ru}(\text{RCOO})_2(\text{diphosphine})$; $\text{Ru}(\text{methylallyl})_2(\text{diphosphine})$; $\text{Ru}(\text{aryl group})\text{X}_2(\text{diphosphine})$; $\text{RuCl}_2(\text{COD})$; $[\text{Rh}(\text{COD})_2]\text{X}$; $\text{RuX}_2(\text{diphosphine})$; $\text{RuCl}_2(=\text{CHR})(\text{PR}'_3)_2$; $\text{Ru}(\text{ArH})\text{Cl}_2$; $\text{Ru}(\text{COD})(\text{methylallyl})_2$; $[\text{Ir}(\text{COD})_2\text{Cl}]_2$; $[\text{Ir}(\text{COD})_2]\text{X}$; $\text{Cu}(\text{OTf})$; $\text{Cu}(\text{OTf})_2$; $\text{Cu}(\text{Ar})\text{X}$; CuX ; NiX_2 ; $\text{Ni}(\text{COD})_2$; $\text{MoO}_2(\text{acac})_2$; $\text{Ti}(\text{OiPr})_4$; $\text{VO}(\text{acac})_2$; MeReO_3 ; MnX_2 or $\text{Mn}(\text{acac})_2$; wherein each R and R' can independently be alkyl or aryl; Ar is an aryl group; and X is a counteranion. The preferred counteranions include halogen, BF_4 , ClO_4 , SbF_6 , CF_3SO_3 and a mixture thereof.

The catalyst may be prepared in situ or as an isolated compound. An example of the preferred catalyst

of the present invention is chiral Ir-f-binaphane catalyst.

In another aspect, the present invention includes a
5 process for preparation of an asymmetric compounds using
the catalysts described above. The process includes the
step of contacting a substrate capable of forming an
asymmetric product by an asymmetric reaction and a
catalyst prepared by contacting a transition metal salt,
10 or a complex thereof, and a ligand selected from ligands
represented by A through K.

Suitable asymmetric reactions is include
hydrogenation, hydride transfer, hydrosilylation,
15 hydroboration, hydrovinylation, hydroformylation,
allylic alkylation, cyclopropanation, Diels-Alder
reaction, Heck reaction, isomerization, Aldol reaction,
Michael addition and epoxidation. Preferably, the
asymmetric reaction is hydrogenation and the substrate
20 to be hydrogenated is an ethylenically unsaturated
compound, imine, ketone, enamine, enamide, and vinyl
ester. Suitable catalysts for the hydrogenation of
imines to produce a chiral amine include Ir complex of
chiral f-binaphane and Rh complex of chiral binaphane.

25

To test synthetic utility of (R,R)-binaphane,
asymmetric hydrogenation of enamides using a Rh-(R,R)-
binaphane complex as the catalyst was conducted. The
imine reduction may be carried out in the presence of
30 additional catalysts, such as iodine or H⁺.

Initially, several experiments were performed to screen optimal conditions for hydrogenation of *N*-acetyl-phenylethenamine. $\text{Rh}(\text{COD})_2\text{PF}_6$ was found as a better catalyst precursor compared with a neutral Rh species
5 $[\text{Rh}(\text{COD})\text{Cl}]_2$. Increase of H_2 pressure results in decrease of enantioselectivity of the asymmetric hydrogenation. For example, 85%ee was obtained under 300 psi H_2 while 90%ee was achieved under 20 psi H_2 . Variation of solvents causes dramatic changes in both
10 enantioselectivity and reactivity. While hydrogenation was complete in CH_2Cl_2 with 90%ee, both reactivity and enantioselectivity were lower in methanol (14%ee and 86% conversion).

15 Several enamides were prepared according to literature procedures and were used as substrates for the asymmetric hydrogenation reaction. Table 1 lists results obtained under the optimal conditions for hydrogenation of *N*-acetyl-phenylethenamine (6a).
20 Although very good enantioselectivity has been obtained for hydrogenation of α -arylenamides without substituents in the β -position, the highlight of the Rh-binaphane catalyst is its ability to reduce β -substituted- α -arylenamides with excellent
25 enantioselectivities. Several β -substituted- α -arylenamides with the mixture of (*E*)/(*Z*) isomers were reduced in high enantioselectivities (entries 7-13, 95%-99.6%ee). A small electronic effect was observed. These enantioselectivities with the Rh-(*R,R*)-binaphane
30 catalyst is the highest reported to date. Since the product 7 can be easily converted to the corresponding

arylalkylamine through hydrolysis under acidic condition, hydrogenation with the Rh-binaphane complex provides a practical method for preparing a variety of chiral arylalkylamines.

5

EXAMPLES

General Procedures:

10

All reactions and manipulations were performed in a nitrogen-filled glove box or using standard Schlenk techniques. THF and toluene were dried and distilled from sodium-benzophenone ketyl under nitrogen.

15 Methylene chloride was distilled from CaH_2 . Methanol was distilled from Mg under nitrogen. (R,R)-BDNPB was made a solution of 10mg/ml in toluene before use. Column chromatography was performed using EM silica gel 60 (230~400 mesh). ^1H , ^{13}C and ^{31}P NMR were recorded on
20 Bruker WP-200, AM-300, and AMX-360 spectrometers. Chemical shifts were reported in ppm down field from tetramethylsilane with the solvent resonance as the internal standard. Optical rotation was obtained on a Perkin-Elmer 241 polarimeter. MS spectra were recorded
25 on a KRATOS mass spectrometer MS 9/50 for LR-EI and HR-EI. GC analysis was carried on Helwett-Packard 6890 gas chromatography using chiral capillary columns. HPLC analysis was carried on WatersTM 600 chromatography. Imines were prepared according to a reported procedure.

30

Synthesis of 1,2-Bis((R)-4,5-dihydro-3H-dinaphtho[1,2-c;2',1'-e]phosphepino)benzene:

(R)-2,2'-bistriflate-1,1'-binaphthyl (2):

5

To a solution of (R)-BINOL (40.3 g, 140.7 mmol) in 900 mL of CH₂Cl₂ was added pyridine (40 mL) and followed by dropwise addition of triflic anhydride (50.5 mL, 300 mmol) at 0°C. The mixture was stirred at room
10 temperature for 6h. After removal of the solvent, the residue was diluted with EtOAc (500 mL) and then washed with 5% aqueous HCl (100 mL), saturated NaHCO₃ (100 mL) and brine (100 mL). The organic layer was dried over anhydrous sodium sulfate, concentrated and passed
15 through a silica gel plug (eluted with CH₂Cl₂) to give the (R)-bistriflate (2) (77 g, 99%).

(R)-2,2'-dimethyl-1,1'-binaphthyl (3):

20 To a solution of (R)-bistriflate (2) (77 g, 140 mmol) and NiCl₂•dppp (3.8 g, 7 mmol) in ether (1000 mL) was added dropwise the methyl magnesium bromide (3.0 M, 280 mL) at 0°C. The reaction mixture was heated to refluxing for 24h. The reaction was quenched by
25 addition of water (200 mL) slowly at 0°C and then diluted with 5% aqueous HCl (200 mL). The aqueous layer was extracted with ether (3 × 100 mL). The combined organic layer was washed with NaHCO₃ (100 mL), dried over anhydrous sodium sulfate and concentrated to afford 3 as
30 light yellow color solid (39.2 g, 99%).

(R)-2,2'-dibromomethyl-1,1'-binaphthyl (4):

A mixture of (R)-2,2'-dimethyl-1,1'-binaphthyl (3) (39.2 g, 138.8 mmol), N-bromosuccinimide (52.4 g, 291.5 mmol) and benzoylperoxide (0.5 g) in tetrachlorocarbon (900 mL) was heated at refluxing and irradiated under a sunlight for three days. The mixture was cooled to room temperature and filtered. The filtrate was concentrated and passed through a silica gel plug. After removal of the solvent, the residue was recrystallized from CH₂Cl₂/hexanes to afford 2,2'-dibromomethyl-1,1'-dinaphthyl (4) (41.1 g, 67.3 %).

(R)-2,2'-dichloromethyl-1,1'-binaphthyl (5):

(R)-2,2'-Dibromomethyl-1,1'-binaphthyl (4, 40 g, 90.8 mmol) and LiCl (30 g, 707 mmol) in DMF (800 mL) was mixed together and stirred at room temperature for 6h. To this mixture was added carefully 5% aqueous HCl (300 mL) (exothermomic reaction occurred). The mixture was then extracted with ether (4 x 400 mL). The organic layer was dried over sodium sulfate, concentrated and recrystallized from CH₂Cl₂/hexane to gave 5 as white solid (30 g, 93%).

(R,R)-1,2-bis{(R)-4,5-dihydro-3H-dinaphtho[1,2-c;2',1'-e]phosphino}benzene:

To a solution of (R)-2,2'-dichloromethyl-1,1'-binaphthyl (5, 0.57 g, 1.62 mmol) and NaH (0.2 g, 8.3 mmol) in THF (20ml) was added 1,2-bis(phosphino)benzene

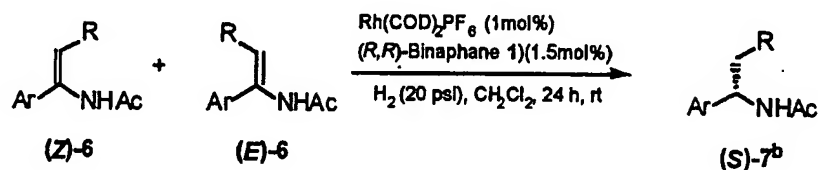
(109 μ l, 0.812 mmol) at -78°C under nitrogen. The mixture was kept stirring at room temperature for 24h and was heated at refluxing for 24 h. After the reaction was completed (monitored by ^{31}P NMR), the solvent was removed via vacuum and the residue was washed with ether (3×15 mL). The organic phase was filtered through a silica gel plug to give the fairly pure product. Further purification by recrystallization from ether afforded binaphane (0.31 g, 55%). ^1H NMR (CDCl_3) 360MHz 7.83-7.8 (4H,m,Ar-H), 7.59-7.56 (2H,m,Ar-H), 7.33-7.15 (16H,m,Ar-H), 7.0-6.9 (2H,m,Ar-H), 6.8-6.7 (2H,m,Ar-H), 6.66-6.63 (2H,d, $J = 8.3\text{Hz}$,Ar-H), 2.97-2.74 (8H, m, ArCH_2); ^{31}P NMR(CDCl_3) -5.63 ppm.

General Procedure for Catalytic Asymmetric Hydrogenation of Enamides:

In a glove box, the Rh-phosphine complex was made *in situ* by mixing $\text{Rh}(\text{COD})_2\text{PF}_6$ (3.7 mg, 0.008 mmol) and binaphane (0.8 mL of 10mg/mL ligand in toluene, 0.012 mmol) in 19.2 mL of CH_2Cl_2 . The mixture was stirred for 30 min. Then 2.5 mL of this solution was transferred to a 10 mL vial with an enamide substrate (0.1 mmol). Hydrogenation was performed at room temperature under 20 psi of hydrogen pressure for 24 h. The hydrogen was released carefully and the reaction mixture was passed through a silica gel plug eluted with EtOAc. The enantiomeric excess was measured by GC or HPLC using a chiral GC or HPLC column without further purification. The absolute configuration of products was determined by

comparing the sign of optical rotation with the reported data (Table 1).

Table 1. Highly Enantioselective Hydrogenation of Enamides Catalyzed by Rh-(*R,R*)-Binaphane Complex ^a



Entry	Substrate ^f	Ar	R	ee % ^d
1	6a	C ₆ H ₅	H	90.0
2	6b	3-Me-C ₆ H ₄	H	89.0
3	6c	4-CF ₃ -C ₆ H ₄	H	82.0
4	6d	4-Ph-C ₆ H ₄	H	75.7
5	6e	4-Cy-C ₆ H ₄	H	90.0
6	6f	2-Np	H	89.5
7	6g	C ₆ H ₅	CH ₃	99.1
8	6h	4-MeO-C ₆ H ₄	CH ₃	99.6
9	6i	4-CF ₃ -C ₆ H ₄	CH ₃	97.0
10	6j	C ₆ H ₅	CH ₂ CH ₃	97.0
11	6k	C ₆ H ₅	CH ₂ Ph	95.0
12	6l	C ₆ H ₅	CH(CH ₃) ₂	97.6
13	6m	2-Np	CH ₃	98.3

^a The reaction was carried out at rt under an initial hydrogen pressure of 20 psi for 24 h. The catalyst was made *in situ* by stirring a solution of Rh(COD)₂PF₆ and (*R,R*)-Binaphane in CH₂Cl₂. [Substrate(0.04M)]: [Rh] : (*R,R*)-Binaphane = 100 : 1 : 1.5. The reaction went in quantitative yield. ^b The configuration of product was determined by comparison of optical rotation with reported data. ^c Enamides 6 were made according to the literature methods. ^d Enantiomeric excesses were determined by chiral GC with a supelco chiral select 1000 column or by Chiral HPLC with a Regis (S,S)-Whelk-o1 column.

Highly enantioselective reduction of C-N double bond has drawn much attention in last few decades. Catalytic asymmetric hydrogenation of acyclic imines has remained among the toughest problems in synthetic chemistry despite several systems being effective in asymmetric hydrogenation of cyclic imines.

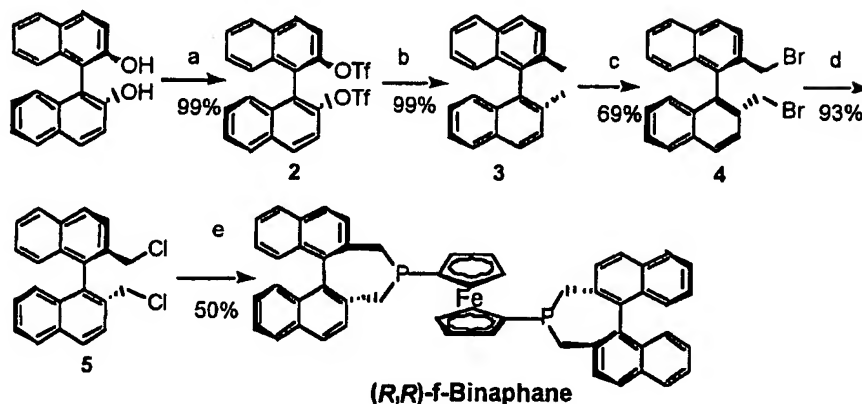
Recently Pfaltz et al. has reported that up to 89%ee can be achieved in enantioselective hydrogenation of N-phenyl imine of acetophenone with an Ir-P-N ligand complex as the catalyst. Enantioselectivities exceeding 90%ee in asymmetric hydrogenation of acyclic imines have rarely been reported in the literature.

According to the present invention, a high enantioselectivity (>99%ee) has been achieved in the synthesis of new chiral ligand (f-binaphane) and in the asymmetric hydrogenation using Ir-f-binaphane complex as the catalyst.

20

Synthesis of Chiral Phosphines (f-Binaphane):

This ligand was made via a route similar to that used to make binaphane. The 1,1'-diphosphinoferrocene was employed as the backbone to afford the new ligand more electron donating property that might be beneficial in transition metal catalyzed asymmetric reactions.

Synthesis of Chiral Phosphine f-Binaphane

^aTf₂O, Py, CH₂Cl₂. ^bMeMgBr, NiCl₂(dppp), Et₂O. ^cNBS, Benzoyl peroxide, CCl₄.
^dLiCl, DMF. ^e1,1'-Diphosphinoferrocene, NaH, THF.

5

(R,R)-1,1'-bis{ (R)-4,5-dihydro-3H-dinaphtho[1,2-c:2',1'-e]phosphino}ferrocene:

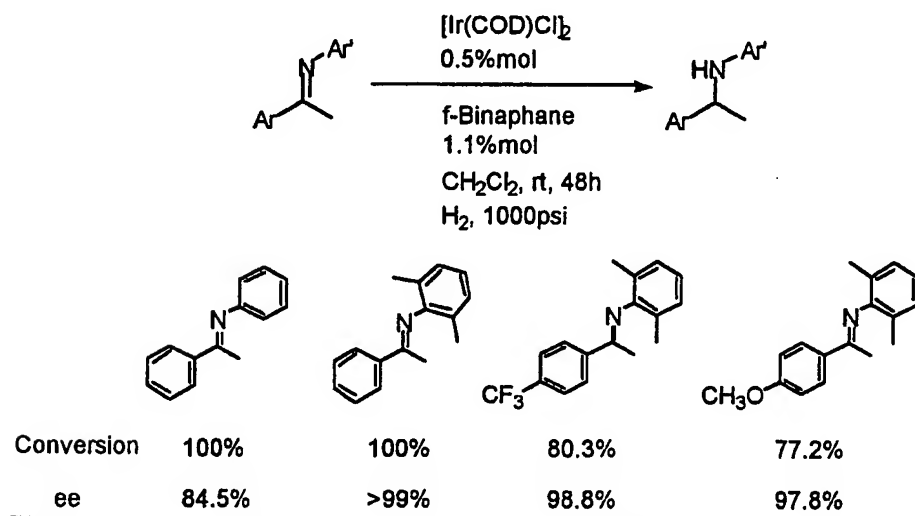
To a solution of (R)-2,2'-dichloromethyl-1,1'-binaphthyl (5, 3.71 g, 10.56 mmol, prepared from (R)-BINOL via (R)-2,2'-bistriflate-1,1'-binaphthyl (2)¹, (R)-2,2'-dimethyl-1,1'-binaphthyl (3)², (R)-2,2'-dibromomethyl-1,1'-binaphthyl (4)³, (R)-2,2'-dichloromethyl-1,1'-binaphthyl (5)⁴ as previously described above) and NaH (2.0 g, 83.0 mmol) in THF (125ml) was added 1,1'-di(phosphino)ferrocene (1.32g, 5.28 mmol) at -78°C under nitrogen. The mixture was kept stirring at room temperature for 24 h and was heated at refluxing for 24 h. After the reaction was completed (monitored by ³¹P NMR), the solvent was

- removed via vacuum and the residue was washed with CH_2Cl_2 (3 x 25 mL). The organic phase was filtered through a silica gel plug to give the fairly pure product. Further purification by recrystallization from
- 5 Hexanes afforded (R,R)-1,1'-bis{(R)-4,5-dihydro-3H-dinaphtho[1,2-c:2',1'-e]phosphepino}ferrocene (2.15 g, 50%). ^1H NMR (CDCl_3) 360MHz δ 7.81-7.76 (8H, m, Ar-H), 7.57-7.54 (4H, m, Ar-H), 7.31-7.13 (10H, m, Ar-H), 7.00-6.90 (2H, m, Ar-H), 6.80-6.70 (2H, m, Ar-H), 6.64-6.62
- 10 (2H, d, J = 8.34 Hz, Ar-H), 2.97-2.74 (8H, m, ArCH_2); ^{13}C NMR (CDCl_3) δ 141.70, 134.71, 134.21, 133.40, 133.27, 132.73, 132.54, 132.34, 131.20, 128.83, 128.69, 128.62, 128.08, 127.80, 127.13, 127.10, 126.32, 125.46, 125.26, 32.50, 29.83; ^{31}P NMR(CDCl_3) δ -6.87. MS m/z : 698 (M^+).
- 15 (References: (1) Uozumi, Y.; Tanahashi, A.; Lee, S-Y.; Hayashi, T. *J. Org. Chem.* 1993, 58, 1945; (2) Sengupta, S.; Leite, M.; Raslan, D. S.; Quesnelle, C.; Snieckus, V. *J. Org. Chem.* 1992, 57, 4066; (3) Maigrot, N.; Mazaleyrat, J-P. *Synthesis*, 1985, 317; and (4) Chong, J.
- 20 M.; MacDonald, G. K.; Park, S. B.; Wilkinson, S. H. *J. Org. Chem.* 1993, 58, 1266).

Enantioselective Hydrogenation of Imines:

- 25 The N-phenyl imine of acetophenone was used to screen the reaction condition. Finally, the optimized condition was set as described below.

Enantioselective Hydrogenation of Acyclic Imines
Catalyzed by Ir-f-Binaphane



5

Other substrates were tried and we found that the bigger N-aryl group plays a very important role for high enantioselectivities. Changing the aryl group of aryl-alkyl ketone has no obvious effect on

10 enantioselectivity.

General Procedure for Catalytic Asymmetric Hydrogenation of Imines:

15 In a glove box, the Ir-phosphine complex was made *in situ* by mixing [Ir(COD)Cl]₂ (6.7mg, 0.01 mmol) and the phosphine complex (0.022 mmol) in 20 mL of CH₂Cl₂ and stirring for 30 min. Then 5 mL of the solution was transferred to a 10mL vial with an imine substrate (0.5

20 mmol). The hydrogenation was performed at rt under 1000

psi of hydrogen pressure for 48 h. The hydrogen was released carefully and the reaction mixture was passed through a silica gel plug eluted with EtOAc. The enantiomeric excess was measured by using GC with a
5 chiral GC column without further purification.

In addition to the chiral ligand binaphane, which has been prepared from chiral BINOL and has shown excellent performance in Rh-catalyzed asymmetric
10 hydrogenation of enamides, the present invention further includes a new chiral chelating phosphine, f-binaphane, which similarly bears binaphthyl moieties but is connected by a 1,1'-bisphosphinoferrocene backbone.

15 The new ligand, f-binaphane, was synthesized by a route similar to the preparation binaphane. Chiral binol was transformed into bistriflates 2, followed by dimethylation through Kumada coupling, bromination with NBS and anion exchanging with LiCl, afforded 1,1'-
20 di(chloromethyl)binaphthyl 5. Refluxing dichloride 5 with 1,1'-bisphosphinoferrocene in the presence of NaH in THF, followed by recrystallization from CH₂Cl₂/Hexanes, gave the pure f-binaphane product as a yellow solid.

25

In addition to the simple synthetic route, the stability of the f-binaphane ligand in the air as well as in solvent (at least for two days) makes the new ligand more attractive.

30

One of the important features of f-binaphane is its electron-rich property due to the two methylene groups and the ferrocene backbone binding to the phosphorus atom. This feature may be responsible for higher
5 asymmetric induction for hydrogenation of unsaturated compounds. Another important feature is the flexibility of the ferrocene backbone that enables the f-binaphane ligand to chelate different transition metals as easily as BINAP does.

10

Furthermore, the coordination of the transition metal with chiral ligand will fix the configuration of the complex and provide a prominent chiral environment in which two binaphthyl groups occupy exactly two
15 opposite quadrants, leaving the other two quadrants vacant. As a result, the ferrocene backbone makes the chiral pocket even broader and deeper and thus accessible to many bulky substrates.

20 The new ligand was applied to the so far unsolved imine reduction problem. In contrast to many catalytic enantioselective hydrogenation of olefins and ketones, only very limited success has been achieved by others in asymmetric hydrogenation of imines. The first attempt
25 of f-Bianphane in hydrogenation of N-(1-phenylethylidene)aniline produced 84%ee at room temperature with 1% Ir-catalyst loading. The optimal condition was screened using the same substrate as the model substrate. The results are summarized in Table 2
30 below.

Table 2. Enantioselective Hydrogenation of N-(1-Phenylethylidene)aniline Catalyzed by Ir-f-Binaphane Complex^a

<div style="text-align: center;"> <p> <chem>c1ccccc1C(=Nc2ccccc2)C</chem> (6a) $\xrightarrow[\text{additives, solvent}]{\text{Ir-f-Binaphane 1\%mol}}$ <chem>c1ccccc1[C@H](Nc2ccccc2)C</chem> (7a) </p> </div>							
entry	additive	solvent	T/°C	P _{H₂} /psi	t/h	conversion%	ee% ^b
1	—	CH ₂ Cl ₂	rt	1000	40	100	84.0
2	—	THF	rt	1000	40	97.2	83.3
3	—	Toluene	rt	1000	40	75.8	23.4
4	—	MeOH	rt	1000	40	71.2	64.0
5	—	CH ₂ Cl ₂	rt	400	48	97.8	84.0
6	—	CH ₂ Cl ₂	-5	1000	64	94.5	84.0
7	Phthalimide 4%	CH ₂ Cl ₂	rt	1000	24	100	82.7
8	ⁿ Bu ₄ NI 4%	CH ₂ Cl ₂	rt	1000	24	100	72.9
9	PhCH ₂ NH ₂ 4%	CH ₂ Cl ₂	rt	1000	24	8.8	62.0
10	AcOH 35eq	CH ₂ Cl ₂	rt	1000	24	48	71.8

^a The reaction was carried out in autoclave by mixing the substrate with catalyst in solvent and pressurized with hydrogen. The catalyst was made in situ by stirring a solution of [Ir(COD)Cl]₂ and (R,R) f-Binaphane in solvent. [Substrate (0.1M)] : [Ir] : (R,R) f-Binaphane = 100:1:1.1. ^b Enantiomeric excess was determined by chiral GC with a Supelco chiral select 1000 column. Absolute configurations was not determined.

Neutral precursor $[\text{Ir}(\text{COD})\text{Cl}]_2$ has better performance than cationic $\text{Ir}(\text{COD})_2\text{PF}_6$. The weakly coordinating solvent CH_2Cl_2 is preferred over other solvent such as THF, toluene or MeOH (entries 1-4).

5 Decreasing the hydrogen pressure has no obvious improvement on enantioselectivity but drops the reactivity (entry 5). The temperature effect is negligible (entry 6). The additive effect that shows a great improvement on enantioselectivity in many systems

10 was not found in this case (entries 7-10). However, phthalimide and tetrabutylammonium iodide did accelerate the reaction to a small extent (entries 7-8).

When 2,6-dimethylphenyl group was employed to

15 replace N-phenyl group, surprisingly >99% enantioselectivity was achieved. To our knowledge, this is the highest enantioselectivity achieved so far for asymmetric hydrogenation of acyclic imines.

20 These results are summarized in Table 3.

Table 3. Enantioselective Hydrogenation of Acyclic Imines Catalyzed by Ir-f-Binaphane Complex

$ \begin{array}{ccc} \text{Ar}-\text{C}(\text{N}-\text{Ar}') & \xrightarrow[\text{H}_2, 1000\text{psi}]{\text{Ir-f-Binaphane 1\%mol, CH}_2\text{Cl}_2, \text{rt}} & \text{Ar}-\text{CH}_2-\text{CH}(\text{N}-\text{Ar}') \\ \text{6(a-j)} & & \text{7(a-j)} \end{array} $						
entry	substrate	Ar	Ar'	t/h	conversion%	ee% ^b
1	6a	Ph	Ph	40	100	84.0
2	6b	Ph	2,6-dimethyl-C ₆ H ₃	44	76.8	>99
3	6c	4-MeO-C ₆ H ₄	2,6-dimethyl-C ₆ H ₃	44	77.2	97.8
4	6d	4-CF ₃ -C ₆ H ₄	2,6-dimethyl-C ₆ H ₃	44	80.3	98.8
5	6e	N-(2,3,4-trihydro-1-naphthylidene) 2,6-dimethylaniline		44	23.8	96.8
6	6f	t-Bu	2,6-dimethyl-C ₆ H ₃	44	14.9	8.3
7	6g	iPr	2,6-dimethyl-C ₆ H ₃	44	28.8	22.6
8	6h	Cy	2,6-dimethyl-C ₆ H ₃	44	24.2	31.4
9	6i	Ph	4-MeO-C ₆ H ₄	12	100	76.6
10	6j	Ph	2-MeO-C ₆ H ₄	14	100	81.4

^a The reaction was carried out at room temperature under an initial hydrogen pressure of 1000psi. The catalyst was made in situ by stirring a solution of [Ir(COD)Cl]₂ and (R,R) f-Binaphane in CH₂Cl₂. [Substrate (0.1M)] : [Ir] : (R,R) f-Binaphane = 100:1:1.1.

^b Enantiomeric excesses were determined by chiral GC with a Supelco chiral select 1000 column. Absolute configurations were not determined.

Changing the electronic property of Ar group has little effect on both reactivity and enantioselectivity (entries 3-4), but replacing the Ar with alkyl group cause a substantial drop of enantioselectivity and reactivity (entries 6-8). The high enantioselectivity (96.8% ee) and low reactivity (23.8% yield) was achieved for the ketamine derived from α -tetralone (entry 5). Replacement of N-Ar' group with methoxyl substituted

phenyl group has great improvement in reactivity without any increment of enantioselectivity (entries 9-10).

This is understandable that the electron donating methoxyl group can destabilize the conjugate system and

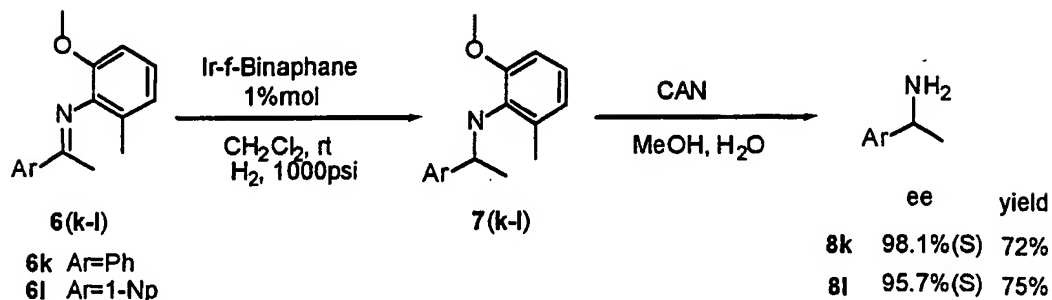
- 5 activated the C=N bond. Base on above facts, the highest enantioselectivity of substrate N-2,6-dimethylphenyl ketamine may be attributable to the steric effect rather than the electronic effect of the dimethyl group.

10

Two substrates (6k and 6l) have been tried under the optimal hydrogenation condition with complete conversions. The hydrogenation products were subjected to deprotection with CAN in MeOH and water at 0°C.

- 15 After workup, the chiral amines were purified by chromatography on silica gel. The enantioselectivities (98.1%ee and 95.7%ee for 8k and 8l respectively) are comparable to their uncleavable analogues. The highly enantioselective hydrogenation of acyclic imines is

20 illustrated below.



Thus, the Ir-f-binaphane complex of the present invention, having a chiral ligand with binaphthyl motif connected by a ferrocene backbone, produces the highest enantioselectivity yet achieved for asymmetric hydrogenation of acyclic imines.

The significance of the new methodology is that it provides a practical method for the synthesis of chiral primary amines.

10

General Procedure for the synthesis of imines 6(a-l):

The ketone (1eq) and the appropriate amine (1eq) were dissolved in dry toluene (40ml) in a flask under nitrogen. The catalytic amount of p-toluenesulfonic acid was added. The flask was equipped with a reflux condenser and a Dean-Stark trap and the mix was heated to reflux for 5hrs. The reaction was quenched by adding saturated NaHCO₃ solution and extracted with ether (3x100ml). The organic layer was combined, washed with brines, dried (Na₂SO₄) and either distilled in vacuo or chromatographed on basic Al₂O₃.

N-(1-Phenylethylidene)aniline (6a): Schnider, P.; Koch, G.; Pretot, R.; Wang, G.; Bohnen, F.-M.; Kruger, C.; Pfaltz, A. *Chem. Eur. J.* 1997, 887.

N-(1-Phenylethylidene)2,6-xylxamine(6b): Schnider, P.; Koch, G.; Pretot, R.; Wang, G.; Bohnen, F.-M.; Kruger, C.; Pfaltz, A. *Chem. Eur. J.* 1997, 887 and Okamoto, H.;

Kato, S.; Ogasawara, M.; Konnai, M.; Takematsu, T.
Agric. Biol. Chem.; EN, 55, 11, 1991, 2733.

N-(1-p-Methoxyphenylidene)2,6-xylxamine (6c): Okamoto,

5 H.; Kato, S.; Ogasawara, M.; Konnai, M.; Takematsu, T.
Agric. Biol. Chem.; EN, 55, 11, 1991, 2733.

N-(1-p-Trifluoromethylphenylidene)2,6-xylxamine (6d):

¹H NMR (CDCl₃, 300MHz), δ8.06-8.04(2H, d, J=6.73Hz, Ar-
10 H), 7.64-7.62(2H, d, J=6.91Hz, Ar-H), 6.99-6.96(2H, d,
J=6.30Hz, Ar-H), 6.87-6.83(1H, t, Ar-H), 2.00(3H, s,
CH₃), 1.93(6H, s, ArCH₃) ¹³C NMR(CDCl₃, 300MHz),
δ164.75, 148.86, 142.49, 133.16, 132.80, 132.44, 132.08,
128.96, 128.38, 127.88, 125.93, 125.87, 125.83, 125.79,
15 125.75, 123.61, 122.94, 119.93, 18.30, 17.91ppm.

N-(2,3,4-Trihydro-1-naphthylidene)2,6-xylxamine (6e):

¹H NMR(CDCl₃, 360MHz) δ8.86-8.84(1H, d, J=7.69Hz, Ar-
H), 7.76-7.66(2H, m, Ar-H), 7.58-7.56(1H, d, J=7.39Hz,
20 Ar-H), 7.43-7.41(2H, d, J=7.45Hz, Ar-H), 7.30-7.26(1H,
t, Ar-H), 3.28-3.24(2H, t, CH₂), 2.64-2.60(2H, t, CH₂),
2.41(6H, s, CH₃), 2.29-2.22(2H, m, CH₂). ¹³C NMR (CDCl₃,
360MHz) δ165.52, 149.61, 141.71, 134.10, 131.17, 129.32,
128.36, 128.06, 126.95, 126.18, 123.12, 30.57, 30.51,
25 23.36, 18.59ppm.

N-(1-Tetrabutylethylidene)2,6-xylxamine (6f): ¹H NMR

(CDCl₃, 360MHz), δ7.22-7.20(2H, d, J=7.56Hz, Ar-H),
7.08-7.04(1H, t, Ar-H), 2.19(3H, s, CH₃), 1.84(3H, s,
30 CH₃), 1.50(6H, s, CH₃). ¹³C NMR (CDCl₃, 360MHz) δ177.29,

149.33, 128.63, 128.28, 127.97, 125.70, 122.93, 122.65,
122.31, 40.98, 28.46, 18.06, 15.69ppm.

N-(1-Isopropylethylidene)2,6-xylxamine (6g): ¹NMR

- 5 (CDCl₃, 360MHz) δ7.00-6.98(2H, d, J=7.52Hz, Ar-H),
6.87-6.83(1H, t, Ar-H), 2.73-2.65(1H, m, CH), 1.99(6H,
s, ArCH₃), 1.59(3H, s, CH₃), 1.25-1.21(6H, d, J=13.60Hz,
CH₃, major isomer (E)), 0.99-0.97(6H, d, J=6.84Hz, CH₃,
minor isomer (Z)) [(E)/(Z)=14:1], ¹³C NMR (CDCl₃, 360MHz)
10 δ175.91, 149.10, 128.23, 126.04, 122.80, 39.22, 20.41,
18.08, 17.93ppm.

N-(1-Cyclohexylethylidene)2,6-xylxamine (6h): ¹HNMR

- (CDCl₃, 360MHz) δ6.89-6.87(2H, d, J=7.46Hz, Ar-H), 6.76-
15 6.72(1H, t, Ar-H), 2.27-2.25(1H, m, CH), 1.88(6H, s,
ArCH₃), 1.92-1.50(5H, m, CH₂), 1.49(3H, s, CH₃), 1.49-
1.22(5H, m, CH₂). ¹³C NMR (CDCl₃, 360MHz) δ175.37,
149.21, 127.88, 125.98, 122.73, 49.41, 30.93, 26.64,
18.60, 18.15ppm.

20

N-(1-Phenylidene)4'-methoxyaniline (6i): Milart, P.;
Sepiol, J. Z. *Naturforsch. B. Anorg. Chem. Org. Chem.*
EN. 41, 3, 1986, 371.

- 25 N-(1-Phenylidene)2'-methoxyaniline (6j): Okamoto, H.;
Kato, S.; Ogasawara, M.; Konnai, M.; Takematsu, T.
Agric. Biol. Chem.; EN, 55, 11, 1991, 2733.

- 30 N-(1-Phenylidene)2'-methoxy-6'-methylaniline (6k): ¹H
NMR (CDCl₃, 360MHz) δ8.39-8.36(2H, m, Ar-H), 7.76-

7.73(3H, m Ar-H), 7.31-7.28(1H, t, Ar-H), 7.18-7.16(1H, d, J=7.57Hz, Ar-H), 7.10-7.08(1H, d, J=8.09Hz), 4.03(3H, s, OCH₃), 2.42(3H, s, CH₃), 2.41(3H, s, CH₃). ¹³C NMR (CDCl₃, 360MHz) δ167.18, 148.68, 139.77, 139.45, 130.81, 5 129.04, 128.91, 128.76, 128.52, 128.29, 127.76, 126.63, 126.30, 124.14, 123.83, 123.02, 109.46, 56.10, 18.29ppm.

N-(1-(1-Naphthyl)ethylidene)2'-methoxy-6'-methylaniline

(61): ¹H NMR (CDCl₃, 360MHz) δ8.65-8.62(1H, d, J=8.49Hz, Ar-H), 7.94-7.92(2H, d, J=8.04Hz, Ar-H), 7.69-7.67(1H, d, J=8.13Hz, Ar-H), 7.62-7.54(3H, m, Ar-H), 7.08-7.06(1H, t, Ar-H), 6.96-6.70(2H, m, Ar-H), 3.93(3H, s, OCH₃), 2.28(3H, s, CH₃), 2.24(3H, s, CH₃). ¹³C NMR(CDCl₃, 360MHz) δ171.60, 148.38, 140.45, 139.34, 15 134.39, 130.93, 129.55, 128.91, 128.75, 128.63, 127.01, 126.40, 125.46, 125.16, 124.01, 123.02, 108.49, 56.30, 23.02, 18.70, 18.40ppm.

General Procedure for Catalytic Asymmetric Hydrogenation
20 of Imines:

The Ir-f-binaphane complex was made *in situ* by mixing [Ir(COD)Cl]₂ (6.7 mg, 0.01 mmol) and f-binaphane (17.7mg, 0.022 mmol) in 20mL of CH₂Cl₂. The mixture was 25 stirred for 30 min. Then 5 mL of this solution was transferred to a 10mL vial with an imine substrate (0.5 mmol). The hydrogenation was performed at rt under 1000 psi of hydrogen. After the reaction, the hydrogen was released carefully and the reaction mixture was passed 30 through a silica gel plug eluted with CH₂Cl₂. The enantiomeric excess was measured by using GC with a

chiral column without further purification. The absolute configuration of products was determined by comparing the retention time with the standard chiral compounds.

5

N-Phenyl-1-phenylethylamine (7a): Schnider, P.; Koch, G.; Pretot, R.; Wang, G.; Bohnen, F.-M.; Kruger, C.; Pfaltz, A. *Chem. Eur. J.* **1997**, 887.

- 10 N-(2',6'-Dimethylphenyl)-1-phenylethylamine (7b): ^1H NMR (CDCl_3 , 360MHz) δ 7.19-7.10(5H, m, Ar-H), 6.85-6.83(2H, d, $J=7.41\text{Hz}$, Ar-H), 6.70-6.66(1H, t, Ar-H), 4.24-4.19(1H, q, CH), 3.09(1H, b, NH), 2.07(6H, s, ArCH₃), 1.42-1.39(3H, d, $J=6.76\text{Hz}$, CH₃). ^{13}C NMR(CDCl_3 , 360MHz) δ 145.75, 145.39, 129.93, 129.33, 128.91, 127.48, 126.61, 122.11, 57.27, 23.13, 19.40ppm.

- N-(2',6'-Dimethylphenyl)-1-p-Methoxyphenylethylamine (7c): ^1H NMR (CDCl_3 , 360MHz) δ 7.35-7.33(2H, d, $J=6.93\text{Hz}$, Ar-H), 7.09-7.07(2H, d, $J=7.51\text{Hz}$, Ar-H), 6.98-6.90(3H, m, Ar-H), 4.44-4.39(1H, q, CH), 3.91(3H, s, OCH₃), 3.30(1H, b, NH), 2.31(6H, s, ArCH₃), 1.63-1.61(3H, d, CH₃). ^{13}C NMR (CDCl_3 , 360MHz), δ 159.00, 145.31, 137.87, 129.91, 129.23, 127.63, 122.02, 114.06, 56.50, 55.83, 22.98, 19.35ppm.

- N-(2',6'-Dimethylphenyl)-1-p-Trifluoromethylphenylethylamine (7d): ^1H NMR (CDCl_3 , 300MHz), δ 7.49-7.46(2H, d, $J=8.24\text{Hz}$, Ar-H), 7.35-7.32(2H, d, $J=8.21\text{Hz}$, Ar-H), 6.89-6.87(2H, d, $J=7.47\text{Hz}$, Ar-H), 6.75-6.70(1H, t, Ar-H). 4.32-4.25(1H, q, CH), 3.11(1H, b, NH), 2.08(6H, s,

ArCH₃), 1.47-1.44 (3H, d, J=6.74Hz, CH₃). ¹³C NMR (CDCl₃, 360MHz), δ144.90, 129.74, 129.37, 126.87, 126.40, 125.82, 122.80, 122.35, 56.98, 23.26, 19.30ppm.

- 5 N-(2',6'-Dimethylphenyl)-2,3,4-trihydro-1-naphthylamine (7e): The ¹NMR spectrum was consistent with 7e.

N-(2',6'-Dimethylphenyl)-1-Tetrabutylethylamine (7f):

- ¹NMR (CDCl₃, 360MHz) δ6.88-6.85 (2H, d, J=7.44Hz, Ar-H),
10 6.69-6.65 (1H, t, Ar-H), 3.08-3.02 (1H, q, CH), 2.82 (1H, b, NH), 2.16 (6H, s, ArCH₃), 0.96 (9H, s, CH₃), 0.79-0.77 (3H, d, J=6.49Hz, CH₃). ¹³C NMR (CDCl₃, 360MHz) δ145.78, 129.46, 129.29, 121.31, 60.29, 35.33, 27.00, 19.73, 15.75ppm.

15

N-(2',6'-Dimethylphenyl)-1-Isopropylethylamine (7g): ¹H

- NMR (CDCl₃, 360MHz) δ6.90-6.88 (2H, d, J=7.46Hz, Ar-H),
6.71-6.67 (1H, t, ArH), 3.14-3.07 (1H, m, CH), 2.90 (1H, b, NH), 2.18 (6H, s, ArCH₃), 2.14-1.66 (1H, m, CH), 0.94-
20 0.86 (9H, m, CH₃). ¹³C NMR (CDCl₃, 360MHz) δ145.57, 129.34, 129.07, 121.30, 57.49, 33.87, 19.91, 19.58, 17.99, 17.22ppm.

N-(2',6'-Dimethylphenyl)-1-Cyclohexylethylamine (7h):

- 25 ¹H NMR (CDCl₃, 360MHz) δ6.90-6.88 (2H, d, J=7.42Hz, Ar-H), 6.72-6.68 (1H, t, Ar-H), 3.08-3.03 (1H, m, CH), 2.91 (1H, b, NH), 2.18 (6H, s, ArCH₃), 1.91-1.55 (5H, m, CH/CH₂), 1.17-1.08 (6H, m, CH₂), 0.89-0.87 (3H, d, J=6.51Hz, CH₃). ¹³C NMR (CDCl₃, 360MHz) δ129.24, 128.22,

121.36, 57.29, 44.52, 30.43, 28.80, 27.12, 27.01, 26.90, 19.57ppm.

N-(p-Methoxyphenyl)-1-phenylethylamine (7i): ^1H NMR

5 (CDCl₃, 360MHz) δ 7.27-7.20(4H, m, Ar-H), 7.13-7.10(1H, t, Ar-H), 6.60-6.58(2H, d, J=6.69Hz, Ar-H), 6.37-6.35(2H, d, J=6.70Hz, Ar-H), 4.33-4.28(1H, m, CH), 3.58(1H, b, NH), 3.57(6H, s, ArCH₃), 1.39-1.37(3H, d, J=6.70Hz, CH₃). ^{13}C NMR (CDCl₃, 360MHz) δ 152.39, 145.95, 10 142.00, 129.09, 127.30, 126.38, 115.23, 115.07, 56.17, 54.74, 25.57ppm.

N-(o-Methoxyphenyl)-1-phenylethylamine (7j): ^1H NMR

(CDCl₃, 360MHz) 7.28-7.26(2H, d, J=8.60Hz, Ar-H), 7.26-7.18(2H, m, Ar-H), 7.10-7.04(1H, t, Ar-H), 6.67-6.58(2H, m, Ar-H), 6.51-6.48(1H, t, Ar-H), 6.26-6.24(1H, d, J=7.75Hz, Ar-H), 4.57(1H, b, NH), 4.40-4.35(1H, m, CH), 3.77(3H, s, OCH₃), 1.46-1.44(3H, d, J=6.76Hz, CH₃). ^{13}C NMR (CDCl₃, 360MHz) δ 147.05, 145.88, 137.62, 129.07, 20 127.26, 126.32, 121.63, 116.83, 111.56, 109.74, 55.88, 53.81, 25.62ppm.

N-(2'-Methoxy-6'-methylphenyl)-1-phenylethylamine (7k):

^1H NMR (CDCl₃, 360MHz) δ 7.22-7.14(4H, m, Ar-H), 7.09-7.07(1H, t, Ar-H), 6.69-6.64(1H, t, Ar-H), 6.59-6.53(2H, m, Ar-H), 4.44-4.39(m, 1H, CH), 3.83(1H, b, NH), 3.63(3H, s, OCH₃), 2.16(3H, s, ArCH₃), 1.39-1.37(3H, d, J=6.73Hz, CH₃). ^{13}C NMR (CDCl₃, 360MHz) δ 151.77, 146.14, 136.24, 129.81, 128.63, 127.31, 126.59, 124.01, 121.44, 30 108.98, 56.67, 56.13, 23.90, 19.33ppm.

N-(2'-Methoxy-6'-methylphenyl)-1-naphthylethylamine

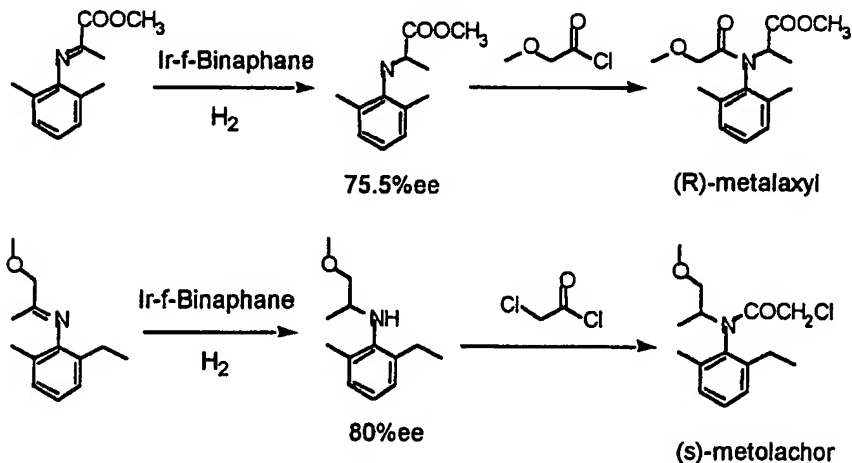
(71): ^1H NMR (CDCl_3 , 300MHz) δ 8.08-8.05(1H, d, $J=8.06\text{Hz}$, Ar-H), 7.74-7.71(1H, d, $J=7.49\text{Hz}$, Ar-H), 7.61-7.59(1H, D, $J=8.16\text{Hz}$, Ar-H), 7.54-7.51(1H, d, $J=6.84\text{Hz}$, Ar-H), 7.38-7.30(3H, m, Ar-H), 6.64-6.54(3H, m, Ar-H), 5.39-5.33(1H, m, CH), 4.08(1H, b, NH), 3.64(3H, s, OCH_3), 2.13(3H, s, ArCH_3), 1.48-1.46(3H, d, $J=6.66\text{Hz}$, CH_3). ^{13}C NMR(CDCl_3 , 300MHz) δ 151.25, 142.46, 136.52, 134.32, 131.42, 129.29, 128.92, 127.62, 126.30, 126.04, 125.76, 124.23, 123.62, 122.86, 120.87, 109.09, 56.17, 52.37, 24.32, 19.61ppm.

General Procedure for deprotection of amines 7(k-l):

The N-protected amine 7k/7l was dissolved in a mixture of MeOH/ H_2O (4:1). CAN (4eq) was added at 0°C , and the mixture was stirred for 6h at the same temperature. Water was added and the solution was washed with CH_2Cl_2 . The aqueous solution was made alkaline by adding 1N NaOH, and then extracted with ethyl acetate. The combined organic layer was washed with brine and dried over Na_2SO_4 . The solvent was removed and the residue was subject to chromatography to afford the pure product 8k/8l.

Another potential application of this transformation is the synthesis of chiral agrochemicals. The preparation of (R)-metalaxyl and (S)-metolachor is shown below.

Synthesis of Agrochemicals via Imine Reduction



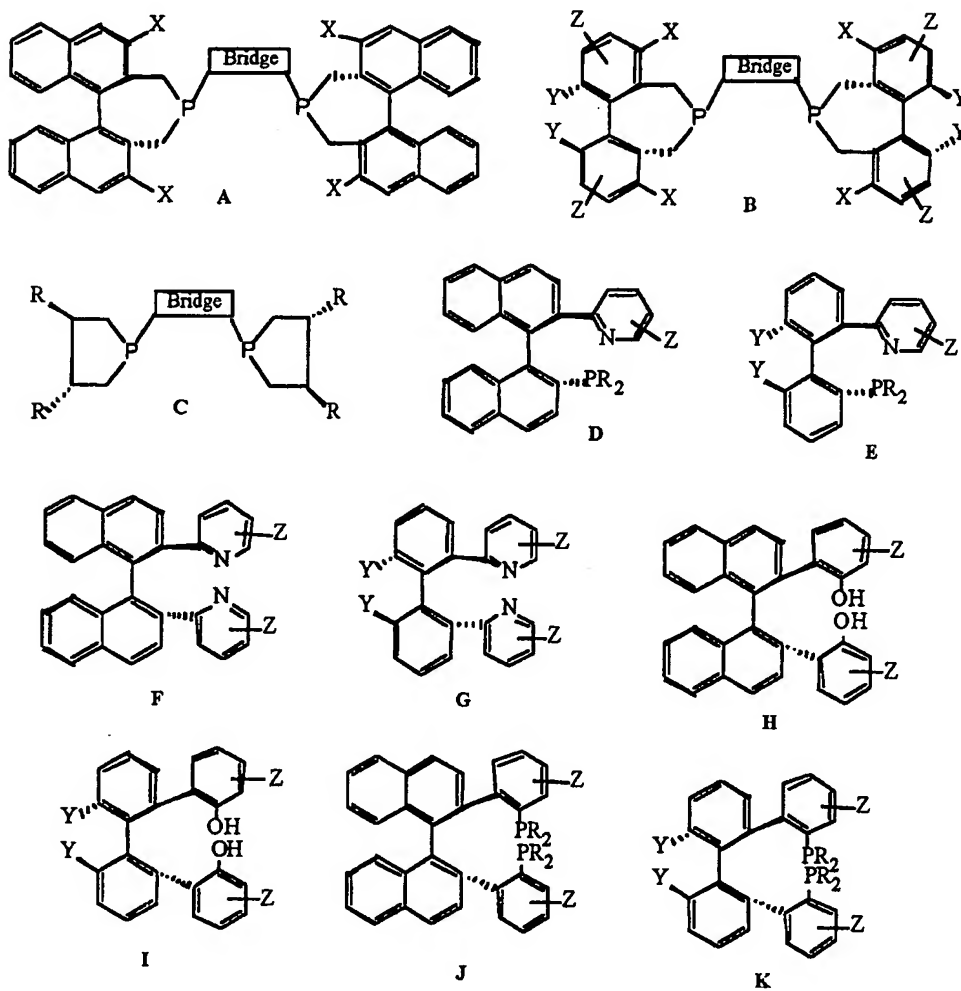
5 Alternative work-up and isolation procedures are also possible, and will be evident to those skilled in the art.

10 The present invention has been described with particular reference to the preferred embodiments. It should be understood that the foregoing descriptions and examples are only illustrative of the invention. Various alternatives and modifications thereof can be devised by those skilled in the art without departing from the spirit and scope of the present invention. 15 Accordingly, the present invention is intended to embrace all such alternatives, modifications, and variations that fall within the scope of the appended claims.

WHAT IS CLAIMED IS:

1. A ligand selected from the group consisting of compounds represented by A through K:

5



wherein the bridge group is selected from the group consisting of: $(CH_2)_n$ wherein n is an integer ranging from 1 to 8, $(CH_2)_nW(CH_2)_m$ wherein n and m are independently an integer ranging from 1 to 8 and W, wherein W is a divalent group selected from the group consisting of: 1,2-divalent phenyl, 2,2'-divalent 1,1'-biphenyl, 2,2'-divalent-1,1'-binaphthyl, ferrocene, and a substituted derivative thereof; wherein each substituent in said substituted derivative is selected from the group consisting of: aryl, alkyl having 1-8 carbon atoms, F, Cl, Br, I, COOR, SO_3R , PR_3R_2 , OR, SR, PR_2 , AsR_2 , SbR_2 , aryloxy, nitro, NR_2 , vinyl, substituted vinyl and a combination thereof, wherein each R is independently selected from the group consisting of: hydrogen, alkyl, aryl, alkaryl and aralkyl; wherein each X is independently selected from the group consisting of: hydrogen, halide, alkyl, aryl, alkoxy, silane, carboxylate and amide; each Y is independently selected from the group consisting of: hydrogen, alkyl, aryl, alkoxy, carboxylate and amide; and each Z is independently selected from the group consisting of: hydrogen, alkyl, aryl, alkoxy, amide, carboxylate, and a heterocyclic compound.

2. The ligand of claim 1, wherein said heterocyclic compound is selected from the group consisting of: a nitrogen heterocycle, a sulfur heterocycle and an oxygen heterocycle.

3. The ligand of claim 1, wherein said ligand is one of the enantiomers.

4. The ligand of claim 1, wherein said ligand is
5 a racemic mixture of enantiomers.

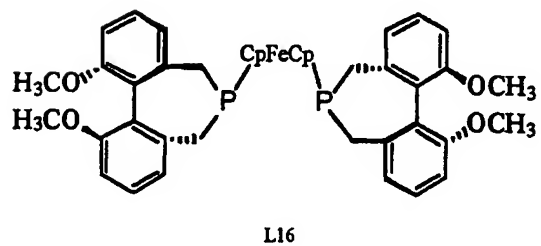
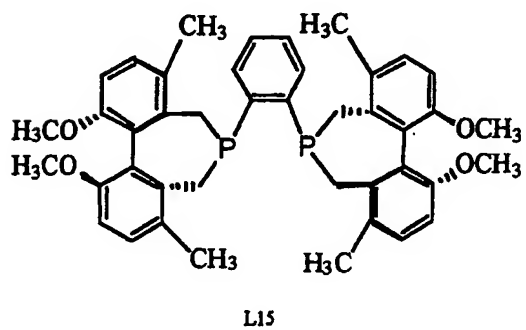
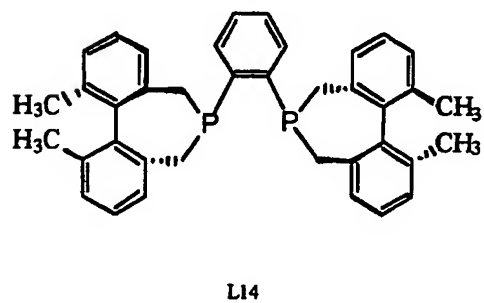
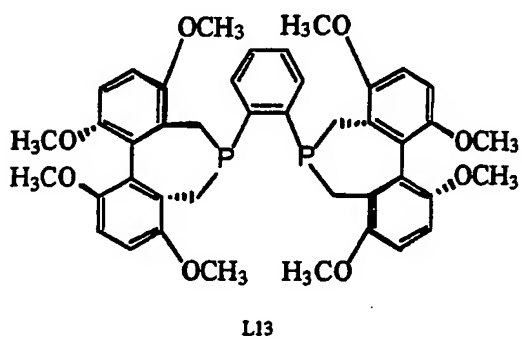
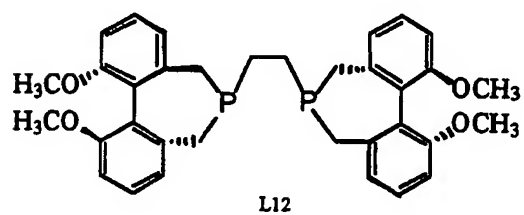
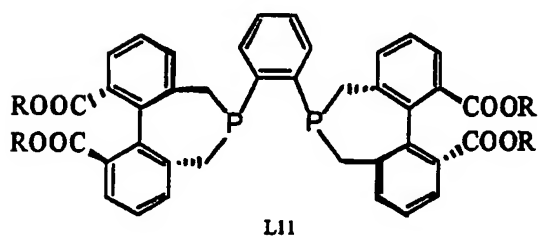
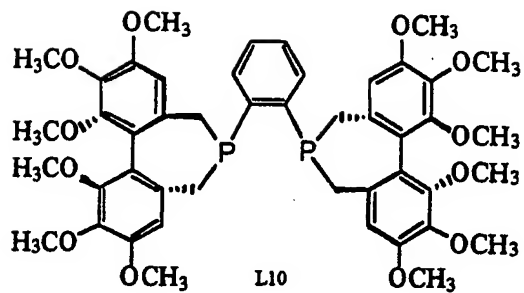
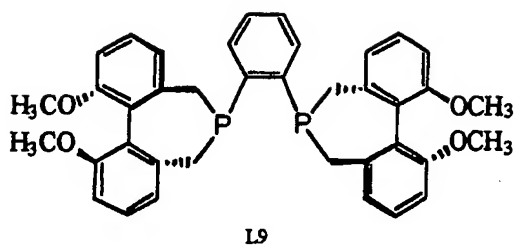
5. The ligand of claim 1, wherein said ligand is a non-racemic mixture of enantiomers.

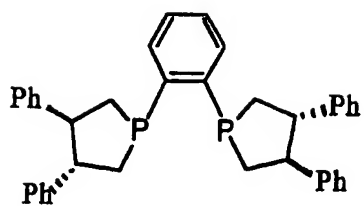
10 6. The ligand of claim 1, having an optical purity of at least 85% ee.

7. The ligand of claim 1, having an optical purity of at least 95% ee.

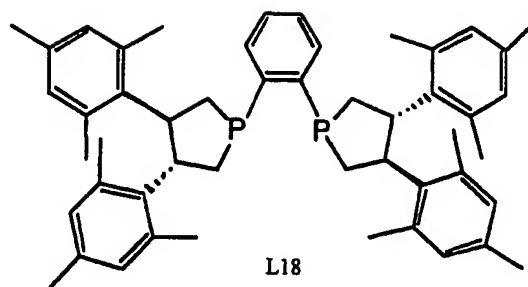
15

8. The ligand of claim 1, wherein said ligand is selected from the group consisting of:

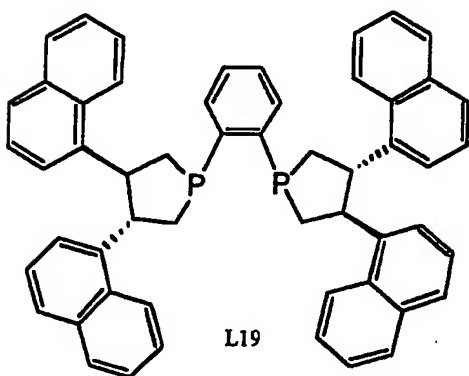




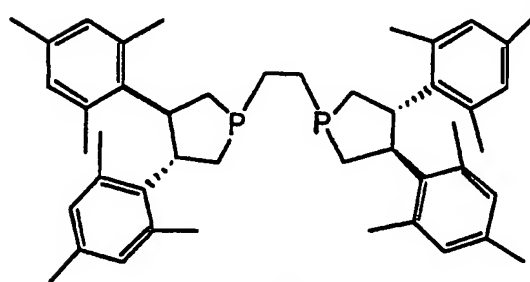
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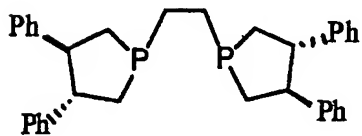
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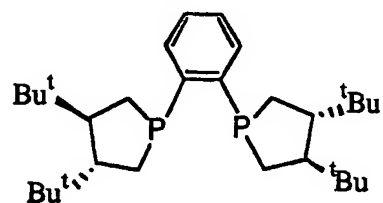
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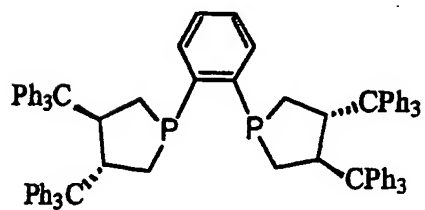
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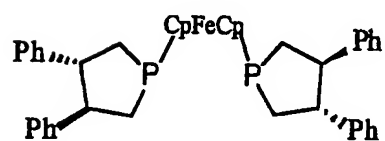
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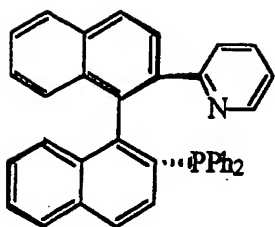


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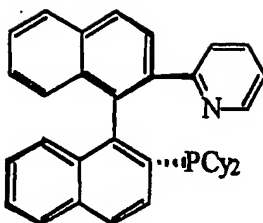


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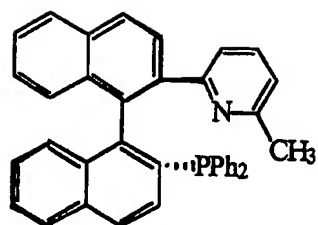
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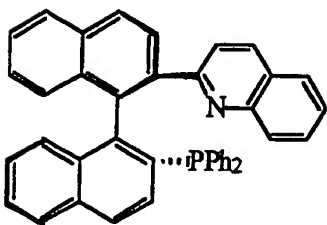
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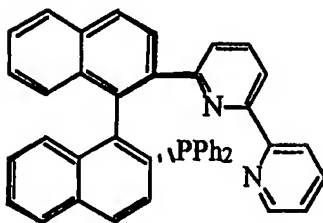
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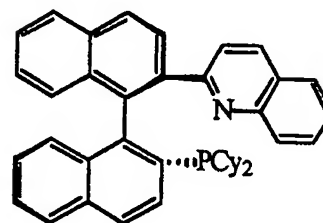
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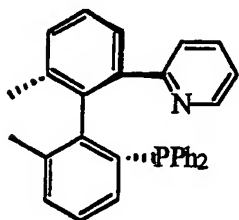
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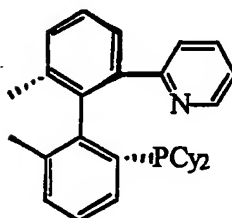
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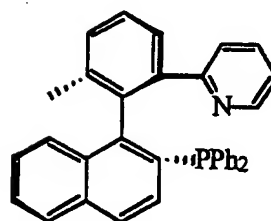
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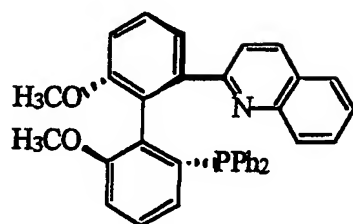
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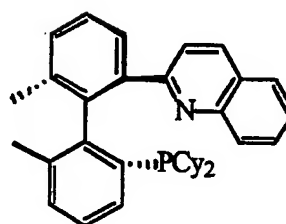
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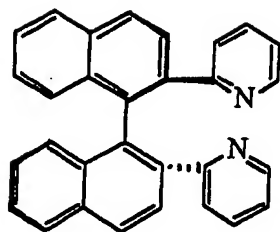
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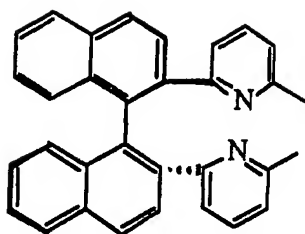
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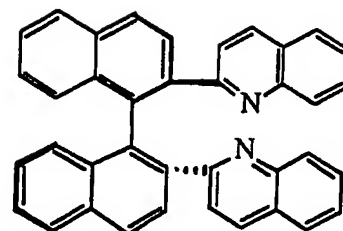
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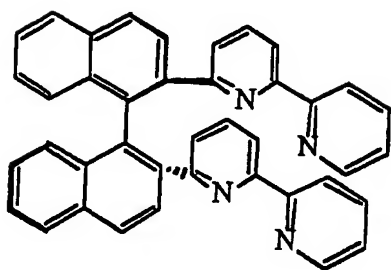
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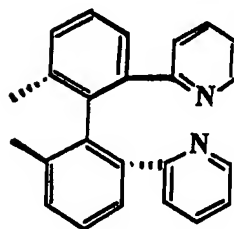
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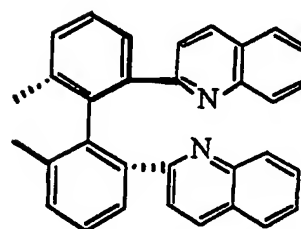
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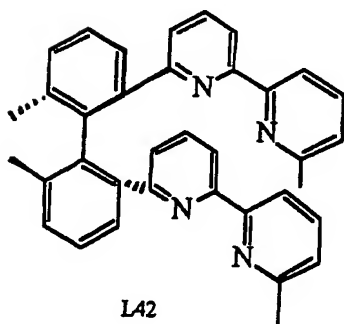
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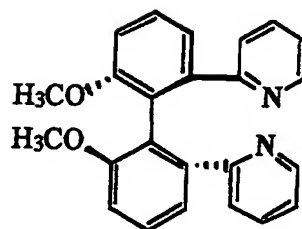
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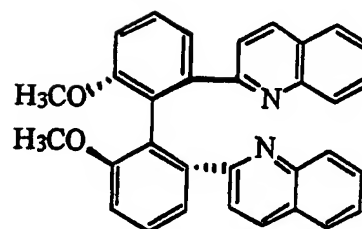
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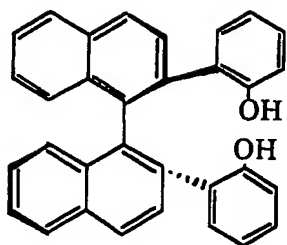
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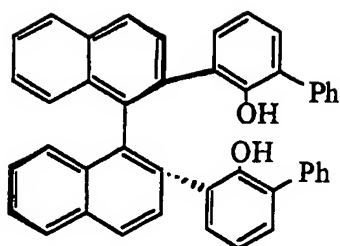
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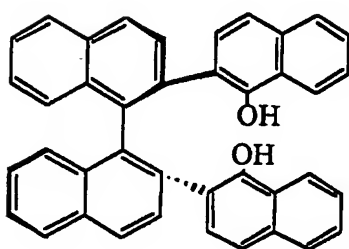
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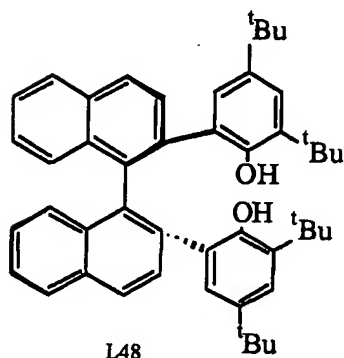
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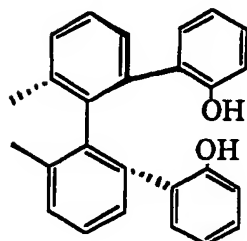
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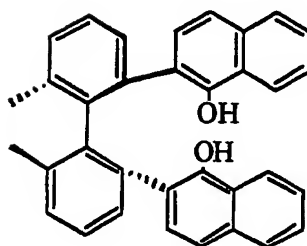
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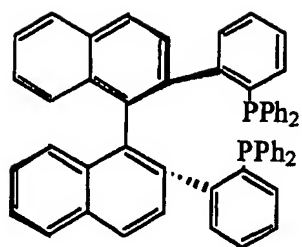
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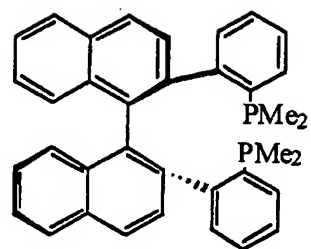
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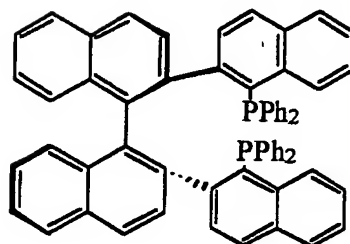
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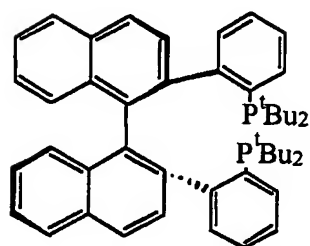
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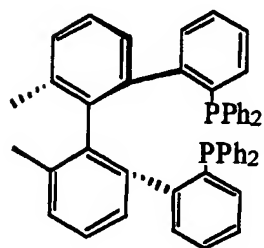
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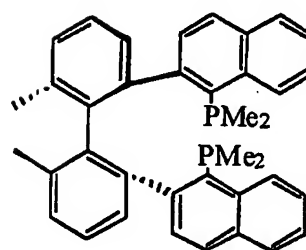
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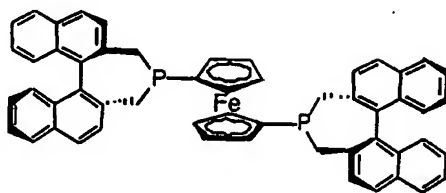
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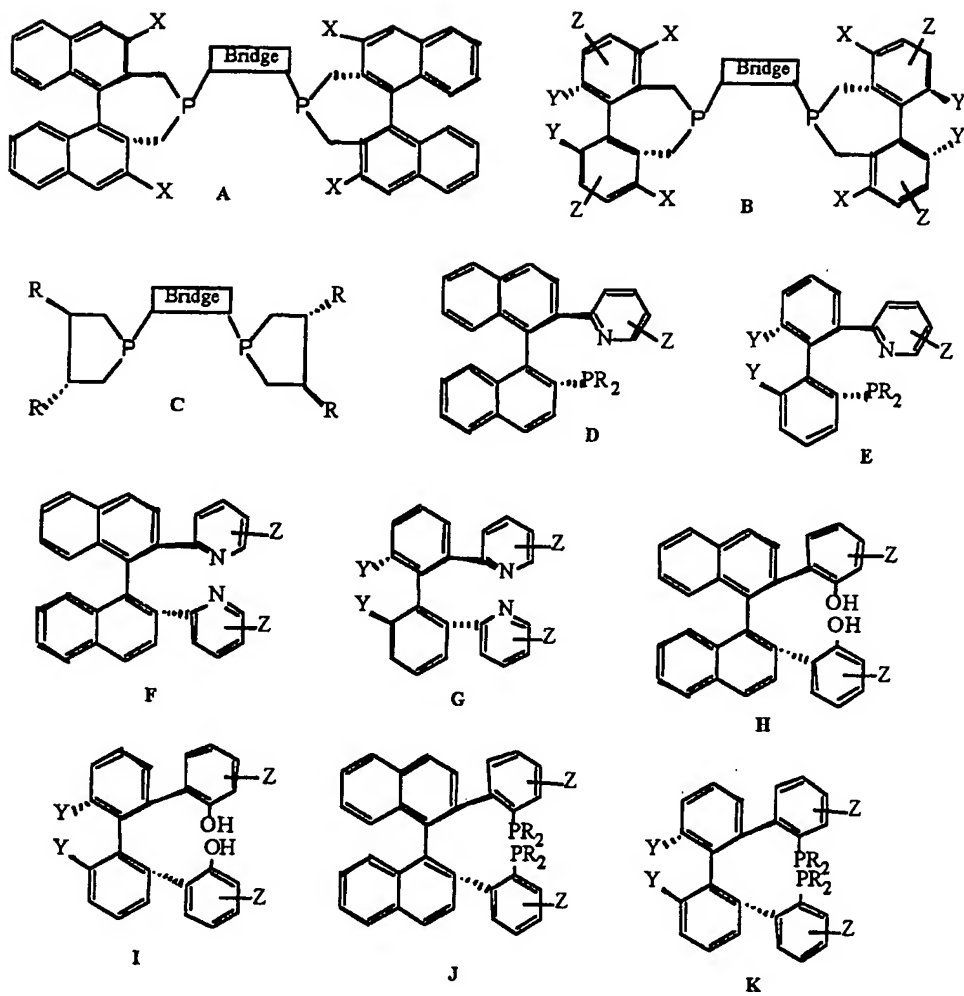
L56

9. The ligand of claim 1, wherein said ligand is a chiral f-binaphane.

10. The ligand of claim 9, wherein said ligand is (*R,R*)-f-binaphane represented by the formula:



11. A catalyst prepared by a process comprising:
contacting a transition metal salt, or a complex
thereof, and a ligand selected from the group consisting
of compounds represented by A through K:



wherein the bridge group is selected from the group consisting of: $(CH_2)_n$ wherein n is an integer ranging from 1 to 8, $(CH_2)_nW(CH_2)_m$ wherein n and m are independently an integer ranging from 1 to 8 and W ,
5 wherein W is a divalent group selected from the group consisting of: 1,2-divalent phenyl, 2,2'-divalent 1,1'-biphenyl, 2,2'-divalent-1,1'-binaphthyl, ferrocene, and a substituted derivative thereof; wherein each substituent in said substituted derivative is selected
10 from the group consisting of: aryl, alkyl having 1-8 carbon atoms, F, Cl, Br, I, COOR, SO₃R, PR₃R₂, OR, SR, PR₂, AsR₂, SbR₂, aryloxyl, nitro, NR₂, vinyl, substituted vinyl and a combination thereof, wherein each R is independently selected from the group consisting of:
15 hydrogen, alkyl, aryl, alkaryl and aralkyl; wherein each X is independently selected from the group consisting of: hydrogen, halide, alkyl, aryl, alkoxy, silane, carboxylate and amide; each Y is independently selected from the group consisting of: hydrogen, alkyl, aryl,
20 alkoxy, carboxylate and amide; and each Z is independently selected from the group consisting of: hydrogen, alkyl, aryl, alkoxy, amide, carboxylate, and a heterocyclic compound.

25 12. The catalyst of claim 11, wherein said catalyst is one of the enantiomers.

13. The catalyst of claim 11, wherein said catalyst is a racemic mixture of enantiomers.

14. The catalyst of claim 11, wherein said catalyst is a non-racemic mixture of enantiomers.

15. The catalyst of claim 11, having an optical
5 purity of at least 95% ee.

16. The catalyst of claim 11, wherein said transition metal is selected from the group consisting of: Pt, Pd, Rh, Ru, Ir, Cu, Ni, Mo, Ti, V, Re and Mn.
10

17. The catalyst of claim 16, wherein said transition metal salt, or complex thereof, is selected from the group consisting of: PtCl_2 ; $\text{Pd}_2(\text{DBA})_3$; $\text{Pd}(\text{OAc})_2$; $\text{PdCl}_2(\text{RCN})_2$; $[\text{Pd}(\text{allyl})\text{Cl}]_2$; $[\text{Rh}(\text{COD})\text{Cl}]_2$; $[\text{Rh}(\text{COD})_2]\text{X}$;
15 $\text{Rh}(\text{acac})(\text{CO})_2$; $\text{Rh}(\text{ethylene})_2(\text{acac})$; $\text{Rh}(\text{CO})_2\text{Cl}_2$; $\text{Ru}(\text{RCOO})_2(\text{diphosphine})$; $\text{Ru}(\text{methylallyl})_2(\text{diphosphine})$; $\text{Ru}(\text{aryl group})\text{X}_2(\text{diphosphine})$; $\text{RuCl}_2(\text{COD})$; $[\text{Rh}(\text{COD})_2]\text{X}$; $\text{RuX}_2(\text{diphosphine})$; $\text{RuCl}_2(=\text{CHR})(\text{PR}'_3)_2$; $\text{Ru}(\text{ArH})\text{Cl}_2$; $\text{Ru}(\text{COD})(\text{methylallyl})_2$; $[\text{Ir}(\text{COD})_2\text{Cl}]_2$; $[\text{Ir}(\text{COD})_2]\text{X}$;
20 $\text{Cu}(\text{OTf})$; $\text{Cu}(\text{OTf})_2$; $\text{Cu}(\text{Ar})\text{X}$; CuX ; NiX_2 ; $\text{Ni}(\text{COD})_2$; $\text{MoO}_2(\text{acac})_2$; $\text{Ti}(\text{OiPr})_4$; $\text{VO}(\text{acac})_2$; MeReO_3 ; MnX_2 and $\text{Mn}(\text{acac})_2$; wherein each R and R' is independently selected from the group consisting of: alkyl or aryl; Ar is an aryl group; and X is a counteranion.
25

18. The catalyst of claim 11, wherein said counteranion X is selected from the group consisting of: halogen, BF_4 , ClO_4 , SbF_6 , CF_3SO_3 and a mixture thereof.

19. The catalyst of claim 11, wherein said ligand is chiral f-binaphane and said transition metal is Ir.
30

20. The catalyst of claim 11, prepared in situ or as an isolated compound.

5 21. A process for preparation of an asymmetric compound comprising:

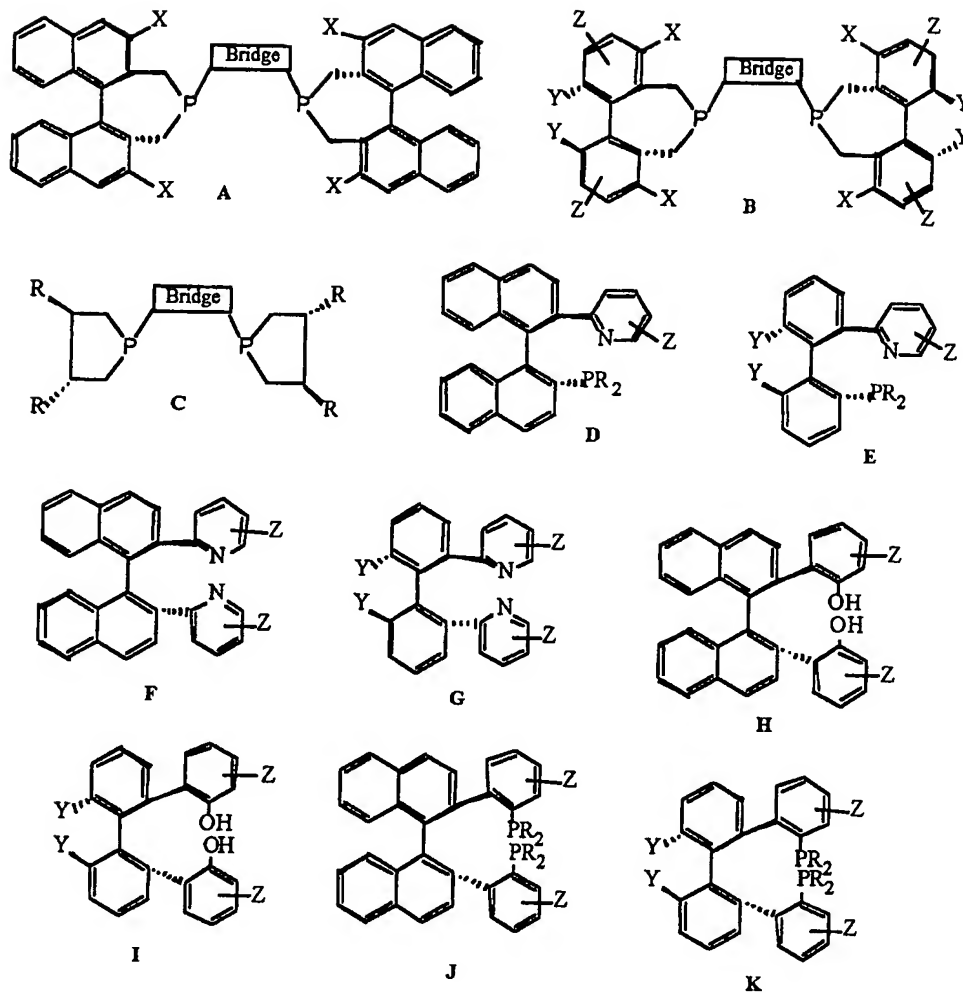
 contacting a substrate capable of forming an asymmetric product by an asymmetric reaction and a catalyst prepared by a process comprising: contacting a
10 transition metal salt, or a complex thereof, and a ligand selected from the group consisting of compounds represented by A through K:

15

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25

30



wherein the bridge group is selected from the group consisting of: $(CH_2)_n$ wherein n is an integer ranging from 1 to 8, $(CH_2)_nW(CH_2)_m$ wherein n and m are

independently an integer ranging from 1 to 8 and W,
wherein W is a divalent group selected from the group
consisting of: 1,2-divalent phenyl, 2,2'-divalent 1,1'-
biphenyl, 2,2'-divalent-1,1'-binaphthyl, ferrocene, and
5 a substituted derivative thereof; wherein each
substituent in said substituted derivative is selected
from the group consisting of: aryl, alkyl having 1-8
carbon atoms, F, Cl, Br, I, COOR, SO₃R, PR₃R₂, OR, SR,
PR₂, AsR₂, SbR₂, aryloxyl, nitro, NR₂, vinyl, substituted
10 vinyl and a combination thereof, wherein each R is
independently selected from the group consisting of:
hydrogen, alkyl, aryl, alkaryl and aralkyl; wherein each
X is independently selected from the group consisting
of: hydrogen, halide, alkyl, aryl, alkoxy, silane,
15 carboxylate and amide; each Y is independently selected
from the group consisting of: hydrogen, alkyl, aryl,
alkoxy, carboxylate and amide; and each Z is
independently selected from the group consisting of:
hydrogen, alkyl, aryl, alkoxy, amide, carboxylate, and a
20 heterocyclic compound.

22. The process of claim 21, wherein said
asymmetric reaction is selected from the group
consisting of: hydrogenation, hydride transfer,
25 hydrosilylation, hydroboration, hydrovinylation,
hydroformylation, allylic alkylation, cyclopropanation,
Diels-Alder reaction, Heck reaction, isomerization,
Aldol reaction, Michael addition and epoxidation.

23. The process of claim 21, wherein said asymmetric reaction is hydrogenation; and said substrate is selected from the group consisting of:
ethylenically unsaturated compound, imine, ketone,
5 enamine, enamide and vinyl ester.

24. The process of claim 21, wherein said catalyst is an Ir complex of chiral f-binaphane and said asymmetric reaction is hydrogenation of imines to
10 produce a chiral amine.

25. A process for preparing a chiral f-binaphane or a substituted derivative thereof, comprising:
contacting chiral 1,1'-di(chloromethyl)binaphthyl
15 or a substituted derivative thereof and 1,1'-bisphosphinoferrocene or a substituted derivative thereof in the presence of a base and a solvent to produce f-binaphane or a substituted derivative thereof.

20 26. The process of claim 25, wherein said base is NaH.

27. The process of claim 25, wherein said solvent is tetrahydrofuran is (THF).

INTERNATIONAL SEARCH REPORT

 International application No.
 PCT/US00/22976

A. CLASSIFICATION OF SUBJECT MATTER

IPC(7) : C07C 39/205, 233/66; C07D 213/04; C07F9/50, 58

US CL : Please See Extra Sheet.

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

U.S. : 546/22, 23, 255, 261, 262, 314, 329; 568/8, 8, 12, 17, 706, 707, 716, 717, 718

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

HAWLEY'S CONDENSED CHEMICAL DICTIONARY-TWELFTH EDITION

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)

Please See Extra Sheet.

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X --- Y	US 5,648,547 A (REGNAT) 15 July 1997 (15.07.97), see columns 1-2 and examples.	1, 3-5, 8 --- 21-22
A --- Y	Database CAPLUS on STN, Chemical abstracts (Columbus, Ohio, USA), CA:119:116906, CHIBA, T. et al, '2,2'-Bis(dicyclohexylphosphino)-6,6'-dimethyl-1,1'-biphenyl(BICHEP)-ruthenium complexes, highly efficient catalysts for asymmetric hydrogenation of carbonyl compounds', Tetrahedron Lett., 34(14) pages 2351-4, August 1993.	11-20, 23 --- 21-22

☒ Further documents are listed in the continuation of Box C.
 ☐ See patent family annex.

* Special categories of cited documents:	*T	later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
A document defining the general state of the art which is not considered to be of particular relevance		
B earlier document published on or after the international filing date	*X*	document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
L document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)	*Y*	document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art
O document referring to an oral disclosure, use, exhibition or other means		
P document published prior to the international filing date but later than the priority date claimed	*A*	document member of the same patent family

Date of the actual completion of the international search

28 OCTOBER 2000

Date of mailing of the international search report

24 NOV 2000

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Facsimile No. (703) 305-3230

Authorized officer

JEAN F VOLLANO

Telephone No. (703) 308-1235

INTERNATIONAL SEARCH REPORT

 International application No.
 PCT/US00/22976

C (Continuation). DOCUMENTS CONSIDERED TO BE RELEVANT		
Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A --- Y	Database CAPLUS on STN, Chemical abstracts (Columbus, Ohio, USA), CA:94:121896, MIYASHITA, A. et al, 'Synthesis of 2,2'-bis(diphenylphosphino)-1,1'-binaphthyl(BINAP), an atropisomeric chiral bis(triaryl)phosphine, and its use in the rhodium(i)-catalyzed asymmetric hydrogenation of alpha-(acylamino)acrylic acids', J. Am. Chem. Soc. 102(27) pages 7932-4, 1980.	11-20 ---- 21-23
A --- Y	Database CAPLUS on STN, Chemical abstracts (Columbus, Ohio, USA), CA:101:171457, MIYASHITA, A. et al, '2-2'-Bis(diphenylphosphino)-1,1'-binaphthyl (BINAP). A new atropisomeric bis(triaryl)phosphine. Synthesis and its use in the rhodium(I)-catalyzed asymmetric hydrogenation of alpha-(acylamino)acrylic acids', Tetrahedron 40(8) pages 1245-53, 1984.	11-20, 24-27 ---- 21-23
A	Database CAPLUS on STN, Chemical abstracts (Columbus, Ohio, USA), CA:130:52214, REETZ, M.T. et al, 'Diphosponites as highly efficient ligands for enantioselective rhodium-catalyzed hydrogenation', Chem. Commun.(Cambridge) (19) pages 2077-2078, 1998.	1-27
A	Database CAPLUS on STN, Chemical abstracts (Columbus, Ohio, USA), CA:118:255441, NANDIN DE CARVALHO, H. et al, '2,2',3,3',5,5'-Hexaphenyl-[1,1'-biphenyl]-4,4'-diols as monomer units for redox polymers: synthesis and polymerization strategies', J. Macromol. Sci., Pure Appl. Chem. A30(6-7), pages 385-97, 1993.	1-8
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INTERNATIONAL SEARCH REPORT

International application No.
PCT/US00/22976

A. CLASSIFICATION OF SUBJECT MATTER:

US CL :

546/22, 23, 255, 261, 262, 314, 329; 568/8, 8, 12, 17, 706, 707, 716, 717, 718

B. FIELDS SEARCHED

Electronic data bases consulted (Name of data base and where practicable terms used):

CAS ONLINE, EAST, BEILSTEIN

search terms: structure drawing for the different groups, phosphopin, binaphthyl, ferrocene, phenol, pyridine, phosphine, enantiomer, hydrovinilation, hydrosilylation, Heck, Michael addition, epoxidation.